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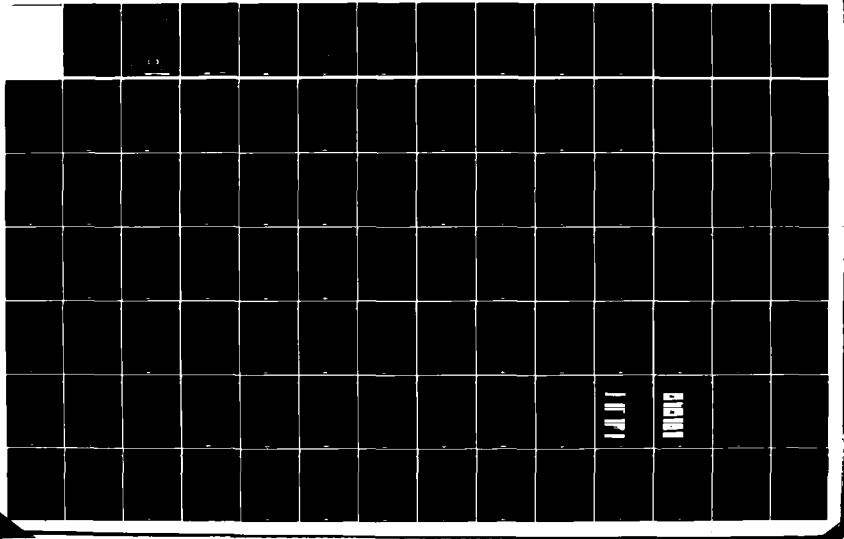

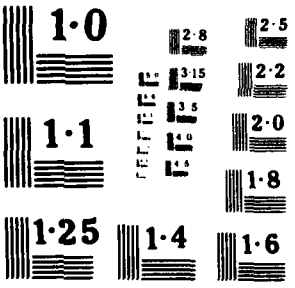
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DTNSRDC-85/003

DAVID W. TAYLOR NAVAL SHIP RESEARCH AND DEVELOPMENT CENTER

Bethesda, Maryland 20084

AD-A151 438

FINGER MATERIALS FOR AIR CUSHION VEHICLES VOLUME I: FLEXIBLE COATINGS FOR FINGER MATERIALS

by

Paul K. Conn and Ivan Snell
Bell Aerospace Textron

and

William Klemens, DTNSRDC

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SHIP MATERIALS ENGINEERING DEPARTMENT
RESEARCH AND DEVELOPMENT REPORT

December 1984

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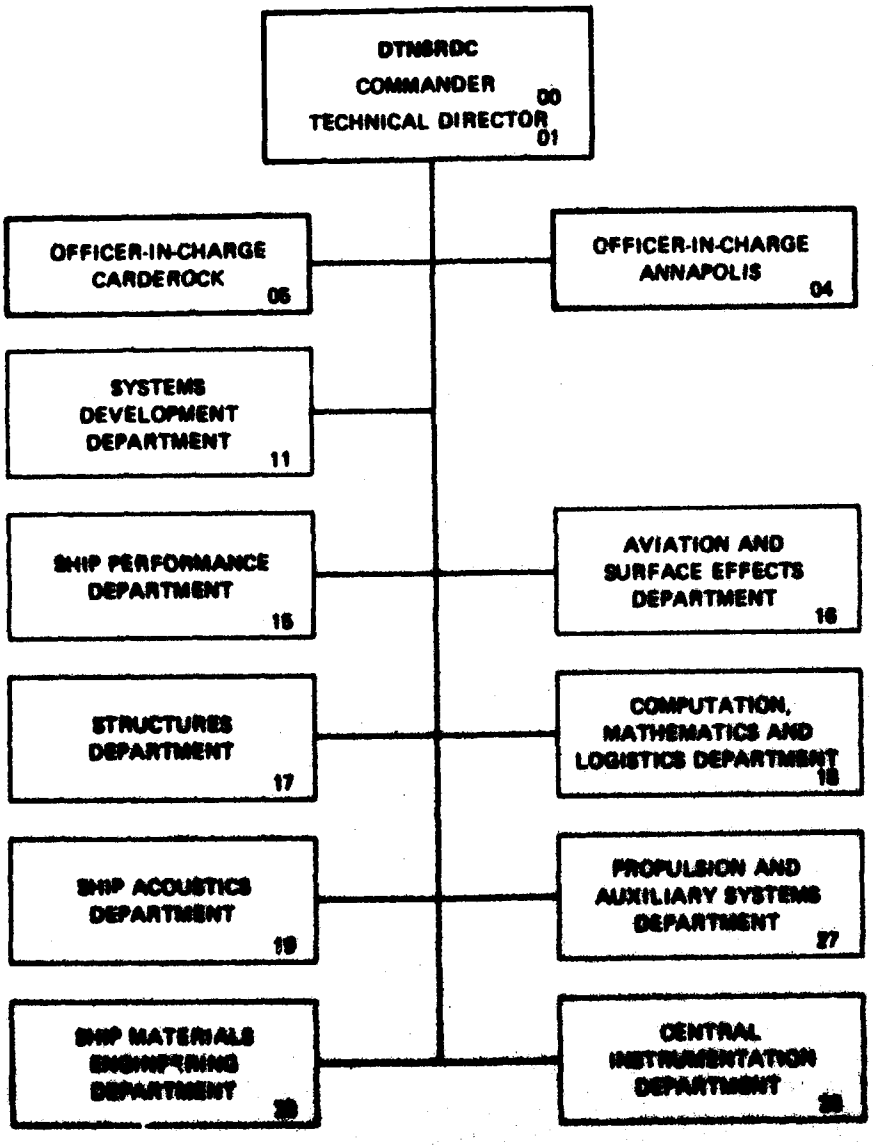

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20 ABSTRACT (Continue on reverse side if necessary and identify by block number) "Twenty polymer formulations from ten selected gum rubber polymers or polymer blends and fourteen formulations of castable liquid polyurethane polymers were characterized as coatings for the coated fabric that is the type material used to make flexible fingers for air cushion vehicles. The formulations were screened for crack (Continued on reverse side)	

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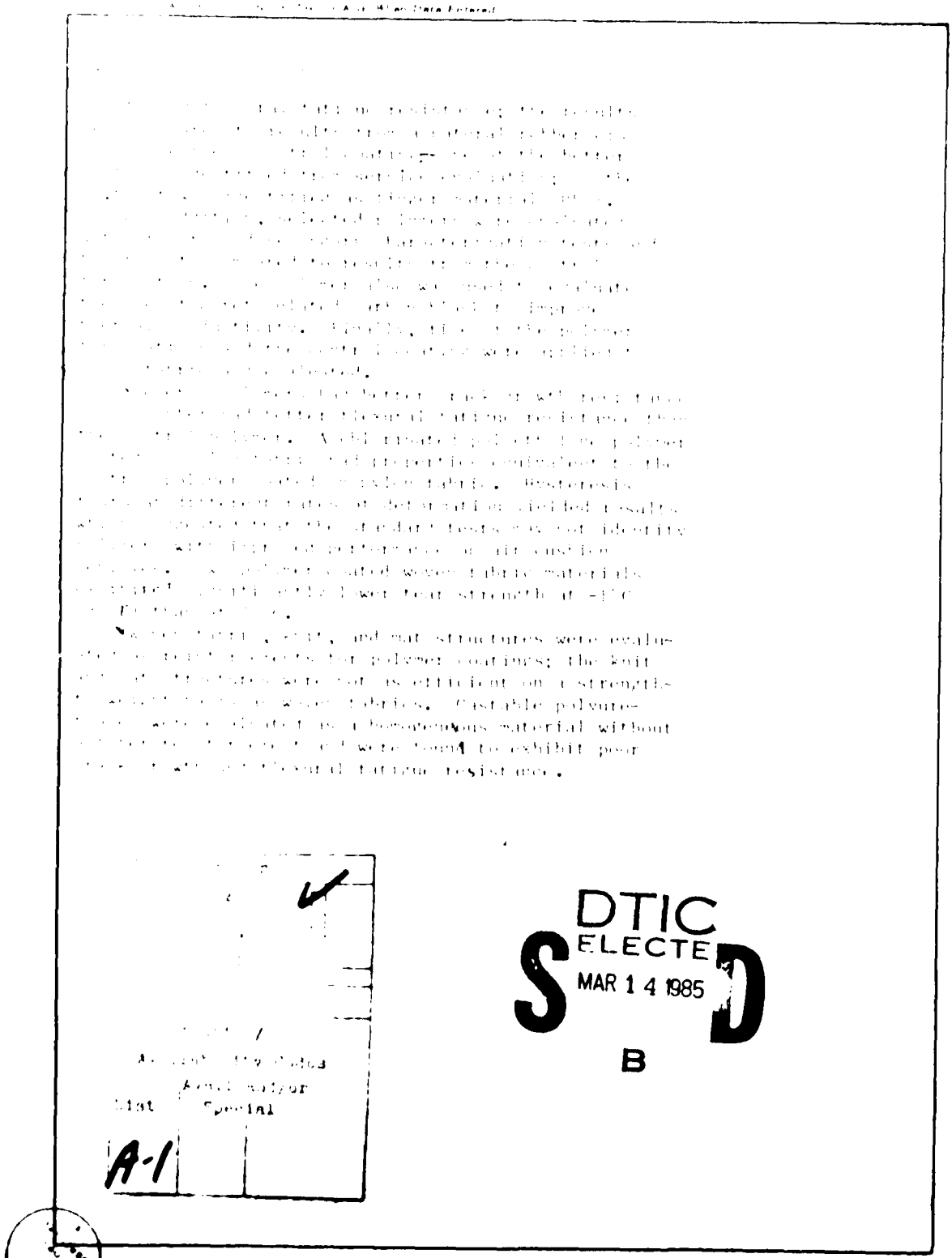

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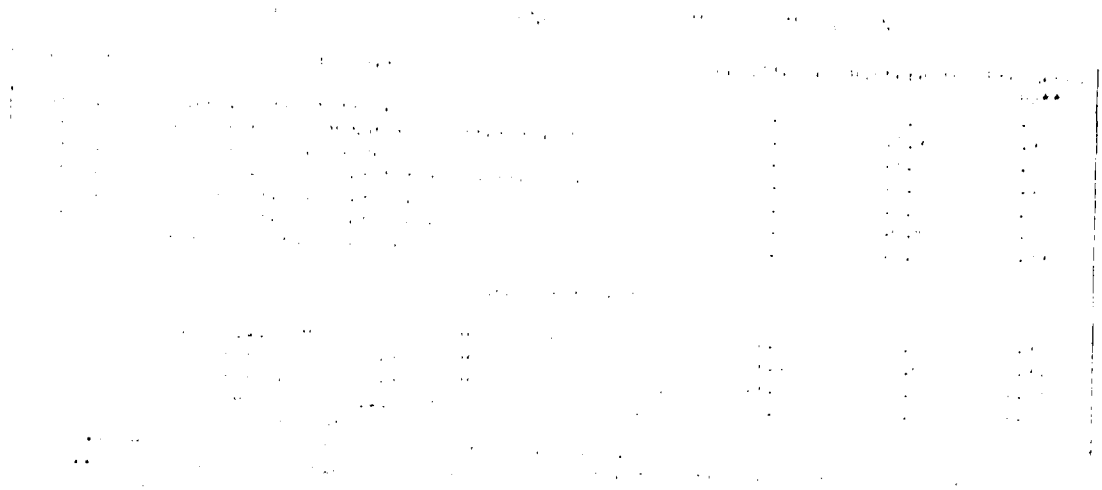
ABBREVIATIONS AND ACRONYMS

ACV	Air Cushion Vehicle
ASTM	American Society for Testing and Materials
BAT	Bell Aerospace Textron
[illegible]	[illegible]
°C	degrees Celsius
cps	cycles per second
cpm	cycles per minute
cm³	cubic centimeters
[illegible]	[illegible]
DTNSRDC	David W. Taylor Naval Ship Research and Development Center
[illegible]	Elastomers and Coatings Branch
EPDM	Ethylene propylene diene terpolymer
°F	degrees Fahrenheit
[illegible]	Fabric Development Laboratory
FEF	Fast Extrusion Furnace Carbon Black
g/cm³	grams per cubic centimeter
[illegible]	[illegible]
GPF	General Purpose Furnace Carbon Black
[illegible]	4,4'-dicyclohexylmethane diisocyanate
[illegible]	[illegible]
[illegible]	Hertz
[illegible]	[illegible]
[illegible]	[illegible] diisocyanate
ISAF-LS	Intermediate Super Abrasion Furnace Carbon Black Low Structure
	Jet Propulsion Fuel No. 5
K	degrees Kelvin
kg	kilogram
LACV-30	Lighter Air Cushion Vehicle - 30 ton payload
[illegible]	Pounds per inch
LSES	Large Surface Effect Ship
MBT	Mercaptobenzothiazole
MBTS	[illegible] disulfide
MDA	Methylene dianiline
MDI	diphenylmethane diisocyanate



of propylene oxide polymer and was ineffective with the chlorinated polyethylene polymer. This is contrary to published results for a natural rubber/SBR butadiene blend.* The concentration used in this study may have been best for the particular polymers.

The cracks growth and flexural fatigue results for the castable liquid polyurethanes are summarized in Table . These formulations were prepared by reacting hydroxyl terminated polyethers, polycarbonates or polybutadienes with a low molecular weight polyol. Specifically, the prepolymers resulting from these reactions were then subsequently cured with either a low molecular weight hydroxy terminated hydrocarbon diol or diamine as discussed in Appendix B. In several instances, a reactive silane was added to the polymer. These particular formulations had suitable viscosity characteristics for casting, and crack growth and flexural fatigue specimens were cast in the appropriate mold immediately after mixing and degassing the liquid. The specimens were cured at room temperature for several days prior to testing; some of the formulations were post cured at 150°F (71°C) for a short period of time.

The results of Table show that all of the formulations had better crack growth resistance than the standard BAT-1, and that FCB-50, PC-120, PC-205, and PC-1205 were substantially better. In contrast, all of the formulations had better flexural fatigue resistance, and two (PC-120 and PC-1205) had approximately the same flexural fatigue resistance as the standard.

On the preliminary solvent/ozone tests, the rubber formulations BAT-1, BAT-2, BAT-3, BAT-4, BAT-5, and BAT-6P, and castable liquid polyurethane formulations FCB-50, PC-120, PC-205, and PC-1205 were selected for characterization by the primary and secondary characterization tests listed in Appendix C.

The results of the primary characterization tests of the polymer formulations are presented in four tables. In Table 5 are summarized the hardness, tensile strength, tear properties, pico abrasion index, and coating adhesion. Values for the dynamic properties yeststests, as measured on the Goodrich flexometer, are presented in Table . Resilience is summarized in Table for the Lupke rebound test and in Table for the Yerzley method. In these tables, the results for the polymers are compared to results for the control formulation, FCB-50.

*Self, G.G., Loban, R.D., "Effects of Molybdenum Disulfide on Elastomer Flex Life and Other Properties," paper presented at a meeting of the Rubber Division, American Chemical Society, New Orleans, LA (7-10 Oct 1975).

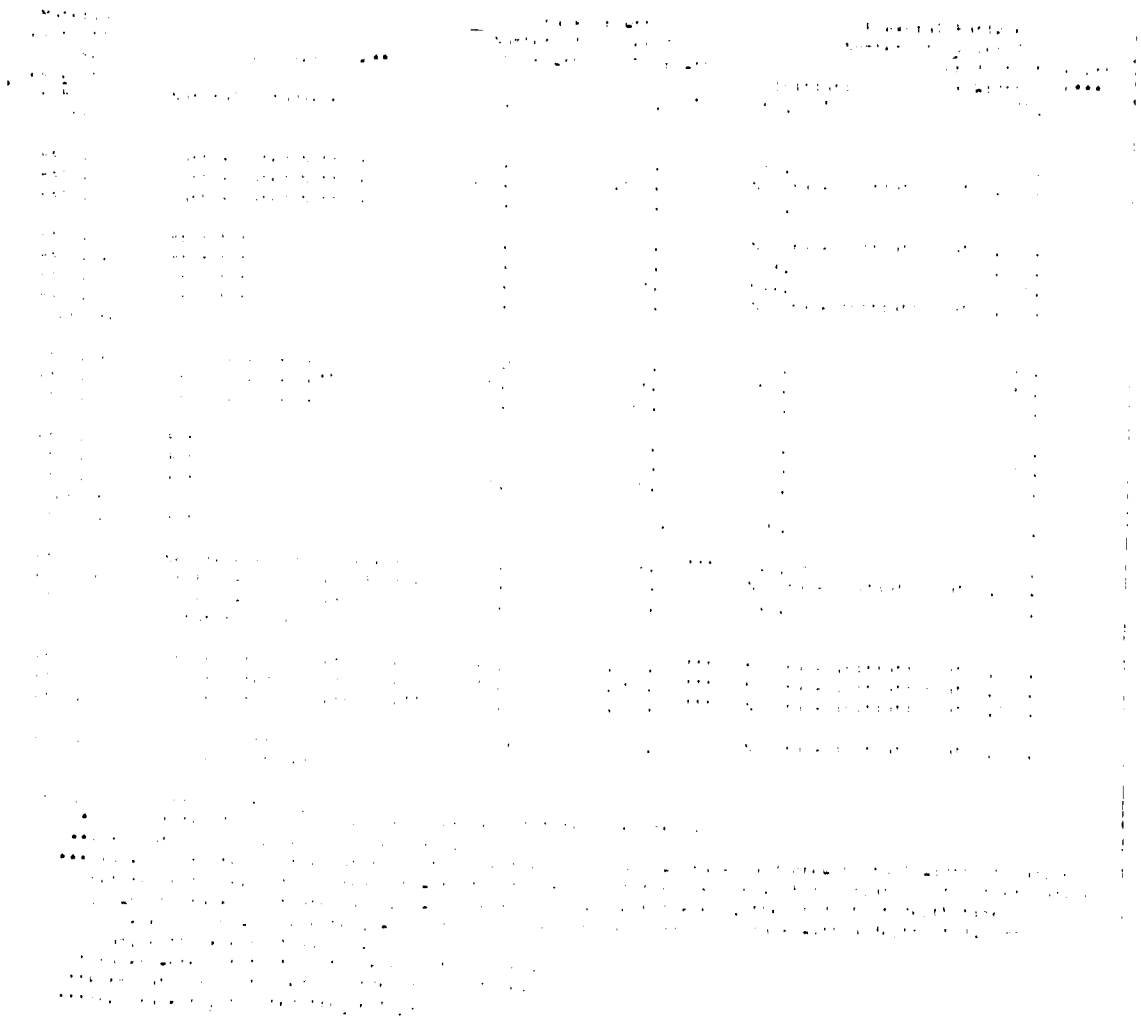
Generally, the variables in the formulations were the type and amount of carbon black, the substitution of some silica for carbon black, the type and amount of [illegible], the type of [illegible] antioxidant, the use of polymer blends, and the type of cure system. [illegible] (BAT- [illegible]) [illegible] (BAT- [illegible]) formulations were [illegible] to investigate the effects of [illegible] [illegible] [illegible] and chlorinated polyethylene (BAT- [illegible]) [illegible] life.

[illegible] flexural fatigue [illegible]

[illegible]

[illegible] formulation [illegible] to the control rubber formulation. [illegible] natural rubber/polybutadiene blend (BAT- [illegible]), [illegible], BAT- [illegible] (SBR block copolymer (NBR blend)), BAT- [illegible] natural blend), BAT- [illegible], BAT- [illegible] and [illegible] chlorinated [illegible] BAT- [illegible] epichlorohydrin fluoroelastomer). Two polymer types, [illegible] rubber, did not yield formulations which were as good as [illegible] properties.

[illegible] results show that the crack growth and the flexural fatigue resistance were [illegible] both [illegible] and [illegible] polymers. However, the use of [illegible] was detrimental to the crack growth and flexural fatigue life



[illegible] evaluated, and the constituents used for castable [illegible] types along with a listing of the charac[illegible], processed data, and a description of the thermal [illegible].

[illegible]

[illegible] selected [illegible] initiation of [illegible] selection and the formulations [illegible] test results, static data were [illegible] with the [illegible] Specification, MIL-P-[illegible]. [illegible] specification [illegible] Specification. [illegible] specification [illegible] The [illegible] Weapons Center [illegible] with the exception [illegible] specification.

[illegible] characteristics. [illegible] were prepared [illegible] formulations were selected [illegible] Finally, two [illegible] rubber coated fabric [illegible] were [illegible] tests.

[illegible] results for the [illegible] rubber [illegible] A more detailed listing of the experimental [illegible] Appendix A. The formulations were developed and prepared [illegible] with the assistance of customer service [illegible] were used at the recommended cure conditions in appropriate [illegible] was used by a commercial company; the cure was based on rheometer data obtained on the mix.

RESULTS OF EXPERIMENTAL WORK

Fingers on air cushion vehicles are subjected to very high frequencies of operation for example, 700 cps at relatively large amplitudes. When rubber coated fabrics are subjected to cyclic stresses through repeated or intermittent flexing, the development of cracks in the coating and delamination from the fabric become the most common causes of failure. During rapid flexing, the ability to sustain rapid strain rates and rapid strain recovery rates without crack formation is important. At high rates of flexing, heat build-up may occur so that high temperature (60-120°C) (140-250°F) properties may be important.

Desirable properties of polymer coatings have been inferred from current knowledge of the seal motions and observed damage. The polymer coating should have high flexural, compressive, and tensile fatigue resistance; high flex crack and crack growth resistance; low flexural modulus; high impact resistance; high abrasion resistance; high tear strength; low hysteresis or excellent high temperature properties; high bond strength to itself and to nylon fabric reinforcement; and resistance to environmental resistance (water, sunlight, ozone, fuel).

The principal criteria for evaluation of the polymers were comparisons of measured properties to the measured properties of known materials which had service experience. In the case of gum rubber polymers, the known material was a natural/polychloroprene rubber blend formulation which was developed by Bell Aerospace Textron on a Navy program and designated BCB-1.2. This formulation was coated on woven nylon fabric and used successfully as bow and side fingers on the British SR.N4 cross channel hovercraft and the Army LACV-30 amphibious air cushion vehicle. Among castable urethanes, the known material was a polyurethane formulation developed by Bell Aerospace Textron and designated BAT-1.2. This formulation was coated on woven nylon fabric and used successfully as side fingers on an Army LACV-60. The woven nylon fabric used with the two different rubber coatings for the LACV-30 fingers had the same weave and configuration, and was designated fabric 128b.* This fabric is a 6 x 6 basket weave of 11 oz/sq yd and was also used as the fabric in this program.

The work is reported under three tasks and five appendices. The experimental results are summarized and discussed in this section. The rationale for polymer

*Woven by Fabric Development Laboratory, Quakertown, PA.

INTRODUCTION

Surface effect ships and air cushion vehicles require a flexible seal or skirt structure to contain the cushion of air on which the craft is supported. The flexibility is necessary to minimize contact (reduce drag) between the vehicle structure and the surface being traversed, which is normally various types of ground terrain or water.

The flexible seals of these craft have typically been fabricated from rubber coated woven fabric laminate materials which were developed originally for applications such as conveyor belts and liquid storage containers. The air cushion craft are exposed to dynamic environments that are more demanding than those for the original applications. These more demanding requirements result from increased mechanical loads, from increased rate of load application, and from environmental extremes which may lead to intolerable or undesirable chemical and physical changes. Hence, substantial improvements of these flexible materials are required to provide adequate structural performance and long operational life with minimum weight and maintenance.

The objective of this program was to develop optimum rubber coatings for flexible seal skirt applications for surface effect ships and vehicles. The primary emphasis was on gum rubber formulations based on selected synthetic rubber polymers and stable liquid polyurethane polymers for use as woven fabric coatings. The tasks included a thermally conductive formulation, formulations for use without reinforcement (homogeneous materials), reinforcements other than woven fabrics, and evaluation of rubber coated fabrics.

Rubber formulations were initially evaluated alone, and then as coatings on a fabric. The evaluations were performed by using standard ASTM tests for selected mechanical properties. Since no correlation had been established between standard mechanical properties and service life on a vehicle, the evaluations were accomplished by comparing results from the experimental materials to results obtained from an existing natural/butadiene rubber blend coated nylon fabric which had been used successfully on an Army LACV-30 air cushion vehicle. Test pieces and sample materials were fabricated for evaluation by Bell Aerospace Textron and the DTNSRDC Annapolis Laboratory.

FOREWORD

The high speed and versatility of Air Cushion Vehicles (ACV) offer significant [illegible] for military applications. However, in amphibious transit over land and water, the lower appendages, called fingers, are subjected to severe flexing and [illegible], shortening the life of the finger. As a result, craft performance is [illegible] life cycle cost increase. The objective of this exploratory development project was to develop sufficient information on fabric and coatings to guide [illegible] selection of coated fabrics as finger materials for Navy air cushion craft.

This volume, the first of three, describes a comprehensive evaluation of [illegible] finger materials for air cushion vehicles. The approach taken, in which materials were screened in preliminary dynamic tests in order to identify [illegible] further research, was a sensible and economic approach. Basically, in this work, the fabric was held constant and the coating was varied. Volume II [illegible] describes the effect from fabric structure on the performance of [illegible] materials. Note, the coating was held constant and the fabric was varied. [illegible] rubber polybutadiene blend coating was common to both endeavors and [illegible] criteria for [illegible] both efforts together. Volume III (to be [illegible]) describes the coated fabric as a finger material. Performance was [illegible] at the [illegible] Aviation Research testing facility called the "Drop [illegible]" which [illegible] operational air cushion vehicles. The information [illegible] Volume II was used to design coated fabric constructions which [illegible] better coated fabrics were tested as detailed in the third volume.

Testing and evaluation performed after completion of the work reported here [illegible] natural rubber polybutadiene blend formulations. This is because [illegible] to offer the greatest resistance to fatigue. Although, this report [illegible] which indicate that natural rubber polybutadiene blend formulations [illegible] offer the greatest resistance to fatigue. No correlation has been [illegible] between those results reported here and the subsequent testing.

[illegible] INFORMATION

[illegible] Surface [illegible] Materials [illegible] Linderfelt, [illegible] Mr. [illegible] Halliwell, [illegible] under [illegible] This report [illegible]

[illegible]

[illegible] Bradshaw, [illegible] Blyth; [illegible] of Direct [illegible] for their assistance [illegible]

[illegible] have made significant contributions. [illegible] Mr. Alfred Watt and Mr. George Pappelt for their support. [illegible] Mr. [illegible] and Mr. [illegible] Powers directed the laboratory work [illegible] characterization, and Mr. Edward Jones assisted in this work. [illegible] advised on the castable liquid [illegible]

[illegible] the overall Project Materials program, [illegible] Air Cushion Program Office [illegible] tested experimental [illegible] the [illegible] Aircraft [illegible] Craft-Experimental Trials Unit [illegible] for trials on the [illegible]; the U.S. [illegible] Fort Belvoir, [illegible] the Canadian [illegible] and Hovérspeed, Ltd. in Dover, Kent, [illegible] A special note of thanks goes to Mr. [illegible] Office of Naval [illegible] Washington, D.C., Dr. H.H. Vanderveldt of the [illegible] Washington, D.C., and Dr. W. Denney Freeston of the [illegible] Institute of Technology, Atlanta, Georgia, for their foresight and guidance throughout this project.

ABSTRACT

Twenty polymer formulations from ten selected gum rubber polymers or polymer blends and fourteen formulations of castable liquid polyurethane polymers were characterized as coatings for the coated fabric that is the type material used to make flexible fingers for air cushion vehicles. The formulations were screened for tear strength and flexural fatigue resistance; the results were compared to results from a natural rubber

In addition, selected polymers were evaluated with ozone and accelerated characterization tests and the results compared to results from the control formulation. A polymer also was used to evaluate the use of a semi-electroconductive carbon black to improve thermal conductivity. Finally, two of the polymer formulations and the control were applied to nylon fabric and evaluated.

Several polymers had better crack growth resistance and a number had better flexural fatigue resistance than the control polymer. A plasticized polyetherurethane polymer coated on nylon fabric had properties equivalent to the control polymer coated on nylon fabric. Hysteresis tests at different rates of deformation yielded results which indicated that the standard tests may not identify polymers with improved performance in air cushion vehicles. Two polymer-coated woven fabric materials exhibited significantly lower tear strengths at -67°C (-88°F) than at 24°C.

Woven fabric, knit, and rib structures were evaluated as reinforcements for polymer coatings; the knit and rib structures were not as efficient as strength members in fabrics as woven fabrics. Castable polymers were evaluated as a homogeneous material without a fabric reinforcement and were found to exhibit poor tear strength and flexural fatigue resistance.

Resilience results on the polymers (Yerzley method) are given in Table 6. In this test, an initial 2" compressive deformation is applied to the specimens; upon release, the specimen begins a train of cycles, which decays due to hysteresis. The results suggest that, in partial agreement with the Lupke results, one chlorinated polyether (BAT 4-1), is at least the equal of the control formulation ECB-1.0; the similarity exists in resilience (% rebound) but not in the speed of rebound, for which BAT-4.5 exhibited a greater speed of recovery. The resilient gum rubber polymers exhibited less resilience than the control with recovery times both faster and slower than the control speed. Two of the polyurethanes (PU-183 and -205) exhibited very low resilience and rates of recovery all greater than the control; the remaining two polyurethanes (PU-195 and -198) straddled the control both in resilience and rate of recovery, although the polymer with the higher resilience had the lower rate of recovery.

In all three of these dynamic tests, polymer deformation is compressive: the Goodrich flexometer is forced high cyclic rate compression-compression, Lupke rebound is a single cycle compression, and the Yerzley rebound is a multiple cycle compressive deformation with the cyclic rate and number of cycles controlled by the viscoelastic properties of the polymer. Since polymer coatings on the seals of air springs and vehicles are subjected to high cyclic rate and concurrent large displacement tension-tension, tension-compression, and compression-compression cyclic stresses, results from the Goodrich flexometer test may be the most important of the three tests. Other cyclic rate, large displacement tension-extension fatigue tests also could have been done but these tests were beyond the scope of the program.

Thermal conductivity of three formulations was measured at each of two different temperatures to evaluate the use of a reticulated carbon black to increase the thermal conductivity of polymer formulations. The three formulations were styrene-ethylene-butylene-styrene block copolymer (SEBS) with 2 parts per hundred parts rubber (pphr) antioxidant 2.0, and 2 pphr Tinuvin 3.0; the same block copolymer formulation with 22 pphr Ketjenblack EC; and the control formulation ECB-0.0.

The thermal conductivity results are presented in Table 7.

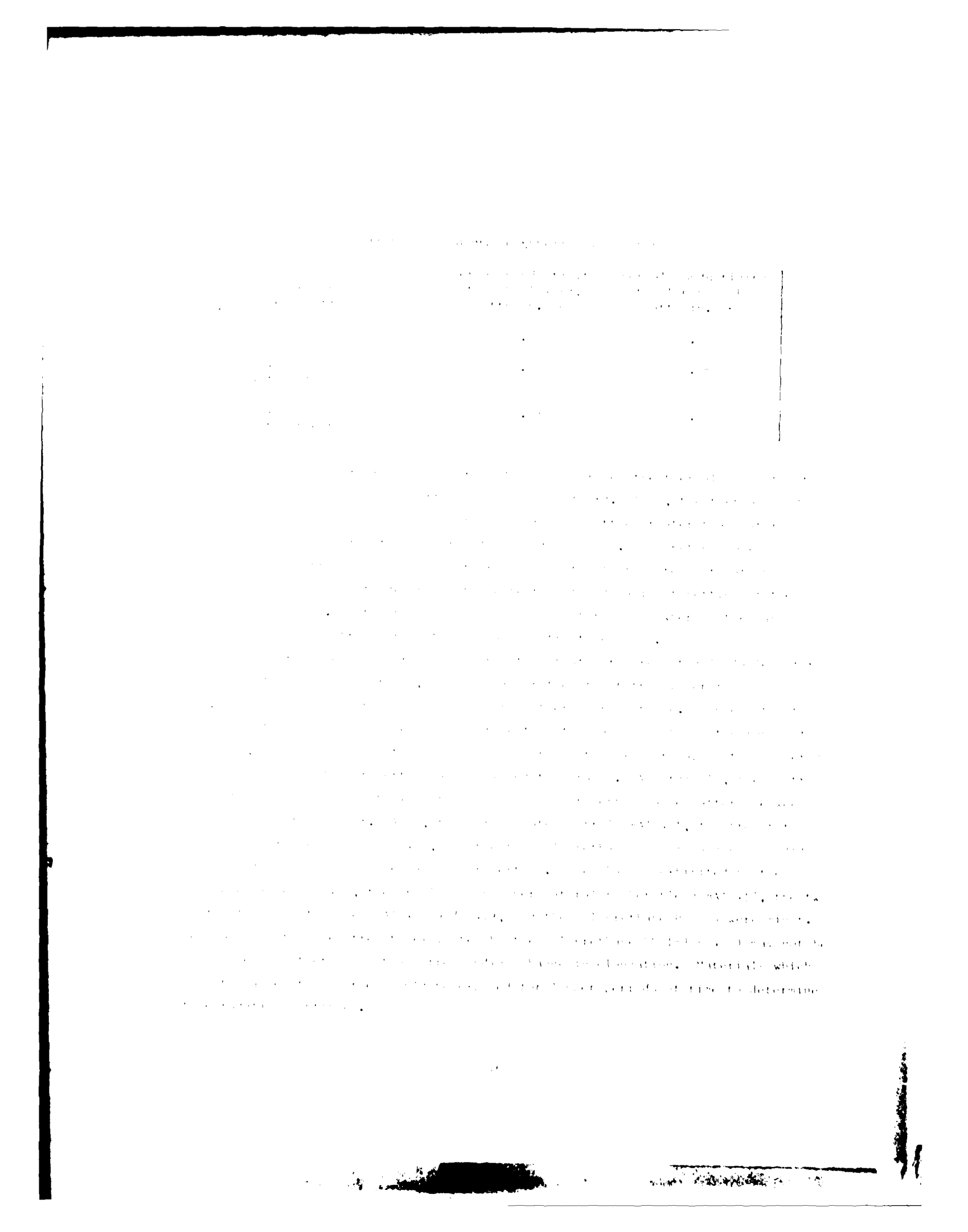


TABLE 10 - OZONE AND OUTDOOR WEATHERING RESISTANCE OF POLYMER FORMULATIONS

Formulation	Polymer	Type of Damage Observed	Time of Damage Days
BAT-[illegible]*	[illegible]ated Nitrile	Shallow, widely spaced [illegible]	[illegible]
BAT-[illegible]*	[illegible] EPDM Natural Rubber Blend	None	[illegible]**
BAT-[illegible]*	[illegible]	Cracks [illegible]	[illegible]
BAT-[illegible]*	[illegible] Natural Rubber Blend	[illegible] cracks	[illegible]
BAT-[illegible]*	[illegible]	[illegible]	[illegible]
BAT-[illegible]*	[illegible]	Cracks, probably [illegible]	[illegible]
[illegible]*	[illegible]	No [illegible]	[illegible]**
		[illegible]	[illegible]
		Extensive [illegible]	
[illegible]***	[illegible]	Cracking [illegible]	[illegible]
[illegible]***	[illegible]	[illegible]	[illegible]
[illegible]***	[illegible]	[illegible]	[illegible]
[illegible]***	[illegible]	[illegible]	[illegible]
[illegible]	[illegible]	[illegible]	[illegible]

[illegible]

*[illegible]

**[illegible]

***[illegible]

[illegible]

The ozone and outdoor weathering resistance of the natural rubber butadiene blend polymer was poor (Table 10), and all the other polymers and polymer blends exhibited better resistance than did the control. The propylene oxide BAT-[illegible], the

[illegible]	Fabric Test Direction	Test Temperature °C (°F)	Ultimate Tensile Strength pli	Ultimate Elongation %	Tear Strength lb	Coating** Adhesion lbs/in.
[illegible]	Warp	49 (120)	2068	38	550	-
		24 (75)	1998	44	420***	
		1 (34)	908	38	29	-
	Fill	49 (120)	844	26	55	
		24 (75)	868	27	50	
		1 (34)	868	19	19	
[illegible]	Warp	49 (120)	1 50	47	[illegible]*	-
		[illegible] (75)	1 [illegible]	29	[illegible]	
		1 (34)	90	17	2	-
	Fill	[illegible] (120)	2 8	26	2	
		[illegible] (75)	2 8	17	5	-
		1 (34)	1	25	11	-
[illegible]	Warp	49 (120)	1 5	[illegible]	[illegible]	
		[illegible] (75)	1 5	[illegible]	[illegible]*	
		[illegible] (34)	11	[illegible]	[illegible]*	
	Fill	49 (120)	1 8	29	[illegible]*	-
		[illegible] (75)	1 6	[illegible]	[illegible]*	
		1 (34)	11	[illegible]	[illegible]*	
[illegible]	Warp	[illegible]	* Lower severity selected from table**			
		[illegible] (120)	1 [illegible]	[illegible]	[illegible]*	
		[illegible]	Test not performed		[illegible]***	-
	Fill	[illegible] (120)	11 [illegible]	29	**	
		[illegible] (75)	12	[illegible]	[illegible]	
		[illegible]	Test not performed		[illegible]***	
[illegible]	Warp	49 (120)	1 5	[illegible]	-	
		[illegible] (75)	11 [illegible]	[illegible]	[illegible]	[illegible]
	Fill	49 (120)	96	25	[illegible]*	-
		[illegible] (75)	110	[illegible]	[illegible]*	
[illegible]	Warp	49 (120)	1080	[illegible]		
		[illegible] (75)	1150	[illegible]	[illegible]*	[illegible]
	Fill	49 (120)	[illegible]	29		
		[illegible] (75)	112	[illegible]	[illegible]*	

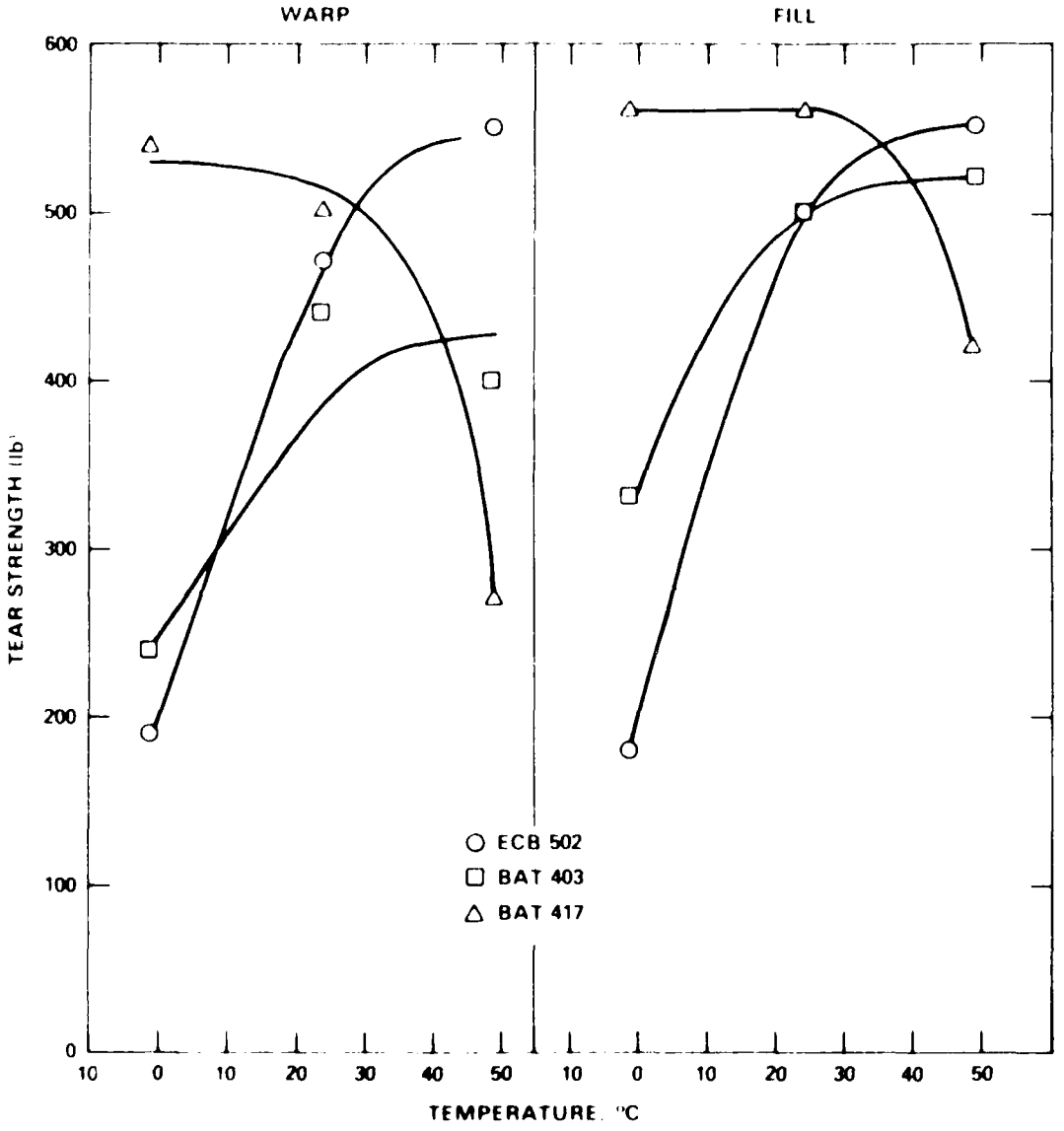
*The nylon woven fabric was a [illegible] x [illegible] Basket Weave of 21 oz/yd² obtained from Fabric Development Laboratory, Hagerstown, MD, with their designation 1785.

**The coat [illegible] was used to bond the gum rubber polymers to the fabric and Chemlok 205 was used as an adhesion promoter with the polyurethane polymers.

***Nylon cord partially pulled through the rubber and partially broke.

*Nylon cord pulled through the rubber and did not break.

**The Neoprene rubber tie coat retarded the cure of the chlorinated polyethylene at the interface so that the chlorinated polyethylene remained a thermoplastic at the interface which softens markedly between 43° (109°F) and 49°C (120°F).


WARP
FILL
600
500
400
300
200
100
0
TEAR STRENGTH (lb)
ECB 502
BAT 403
BAT 417
10
0
10
20
30
40
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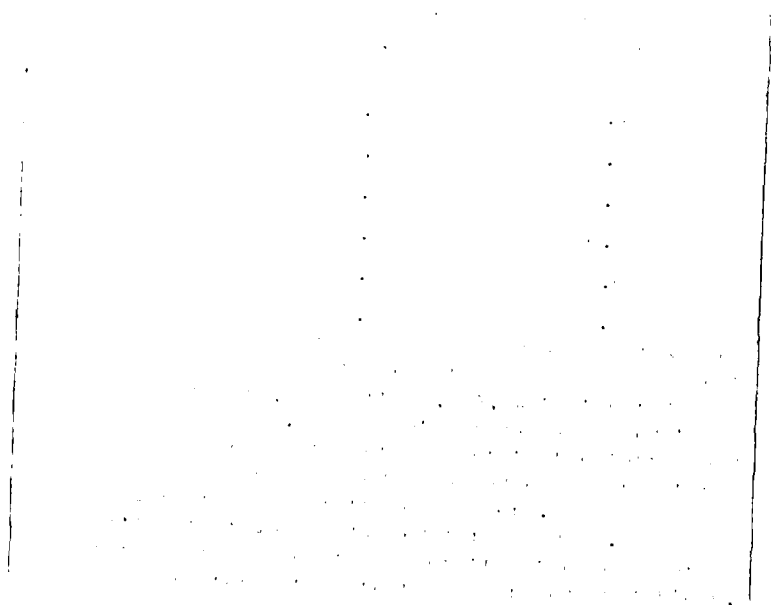
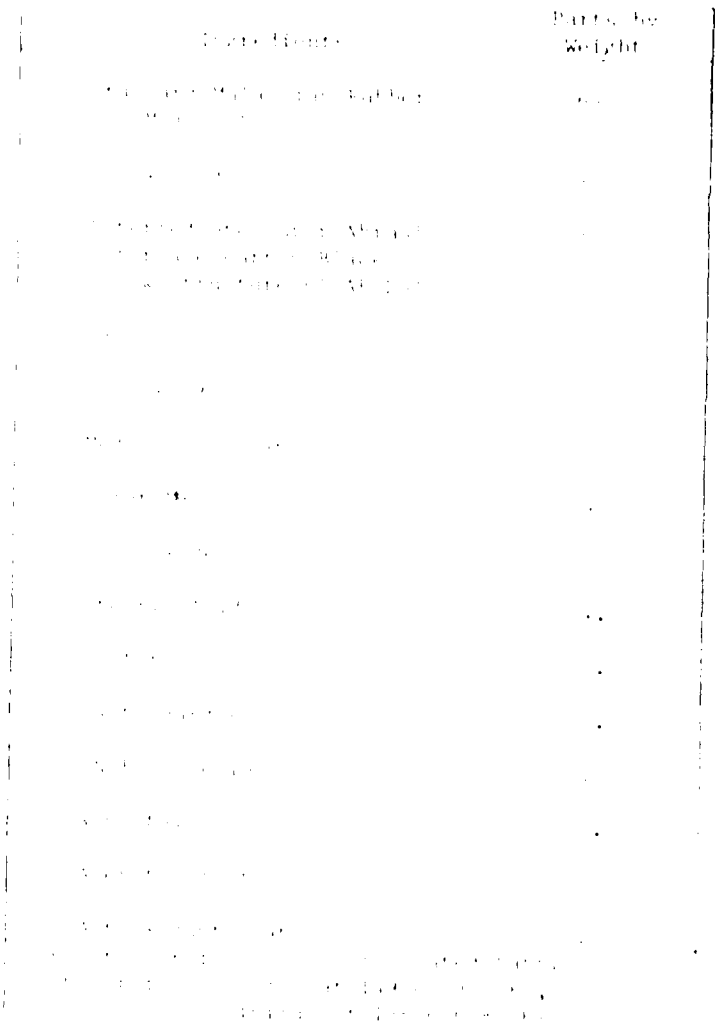
30
40
50
TEMPERATURE, °C





APPENDIX A

SELECTION OF GEM RUBBER POLYMERS

[illegible]

[illegible] polymer formulation development for coating of woven fabric [illegible] was based on [illegible] of two elastomer or polymer blends. [illegible] blends were [illegible] with the exception of the [illegible] were [illegible] polymers. [illegible] were [illegible] of the polymer type [illegible] of [illegible] mechanical properties. The polymers, [illegible] [illegible] were [illegible] [illegible] was investigated with [illegible] [illegible].

[illegible] formulations were [illegible] [illegible] a [illegible] blend polymer which was a [illegible] [illegible] [illegible]. The contents of this formulation are shown [illegible]. This formulation [illegible] [illegible] [illegible] coated fabrics evaluated to date for trusses on the [illegible] [illegible] air cushion vehicle. The formulation [illegible] [illegible] for the Large Surface Effect Ship Program.

[illegible] were evaluated with initial screening tests and [illegible] [illegible] the natural rubber butadiene polymer blend [illegible]. Six test formulations were selected and evaluated further with [illegible] characterization tests (Appendix C) and the results compared [illegible] [illegible] the control polymer formulation. The three best [illegible] were selected, and together with the control formulation, used as [illegible]. The coated fabrics were characterized with property [illegible] (Appendix C).

CONCLUSIONS

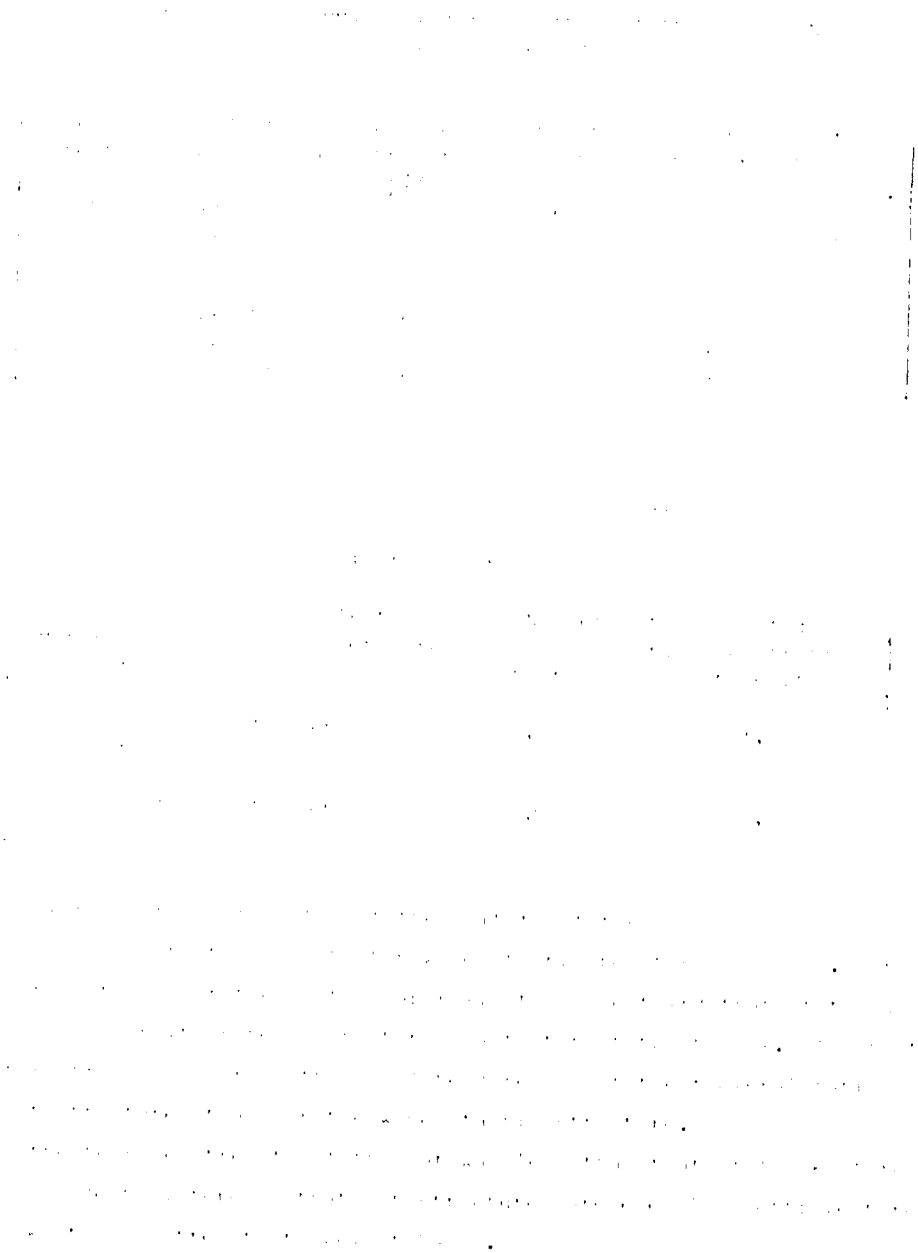
1. Results from laboratory tests at 5 cps on 20 mm rubber formulations prepared from 12 synthetic polymer or polymer blend systems indicated that: a) two of the formulated polymer systems (carboxylated nitrile and chlorinated polyethylene) had tear strength resistance greater than that of the control polymer blend (natural rubber/styrene-butadiene rubber); and b) six of the formulated polymer systems (carboxylated nitrile, butyl, natural rubber/EPDM blend, SB block copolymer/SBR blend, SB block copolymer/natural blend, chlorinated polyethylene, phosphonitrilic fluoroelastomer) had tensile strength resistance greater than that of the control polymer blend.

2. Results from laboratory tests at 5 cps on a variety of liquid castable polyurethane elastomers prepared from five different polyalcohols, four different diisocyanates, and four diamine or dialcohol curatives indicated that: a polyether/MDI/diethylene formulation had better crack growth resistance than the control polyurethane polymer (BAI-Leo); and 80 more formulations had better tensile strength resistance than the control and one formulation had about the same tensile strength resistance as the control.

3. Nylon woven fabric coated with the natural rubber butadiene blend control polymer exhibits a marked reduction in tear strength at -67°F (-55°C) compared to tear strength at room temperature. Similar effects, but to lesser degrees, are exhibited by the terpolymer EPDM/natural rubber blend coated woven fabric at -40°F (-40°C) and the SB block copolymer natural rubber blend coated woven fabric at -20°F (-29°C). The reduction in strength appears to be caused by a modulus change of the polymer coat that the polymer causes the fibers in the yarn bundles to fail more easily. It is recommended that, for complete characterization, polymer coated woven fabrics should be evaluated for mechanical properties over the range of temperatures anticipated for air cushion vehicle use.

4. Conductive SC conductive carbon black appears to more effectively increase the level of conductivity of polymers than regular reinforcing carbon blacks. Since conductive SC is non-reinforcing in most polymers, a detailed study is necessary to correlate with reinforcing carbon blacks to determine if optimum mechanical and thermal properties can be obtained concurrently with improved thermal conductivity.

5. Based on the Pico laboratory abrasion test, the SB block copolymer natural rubber polymer blend had superior abrasion resistance compared to the natural rubber butadiene blend control polymer, and one castable liquid polyurethane polymer exhibited better abrasion resistance than the control.



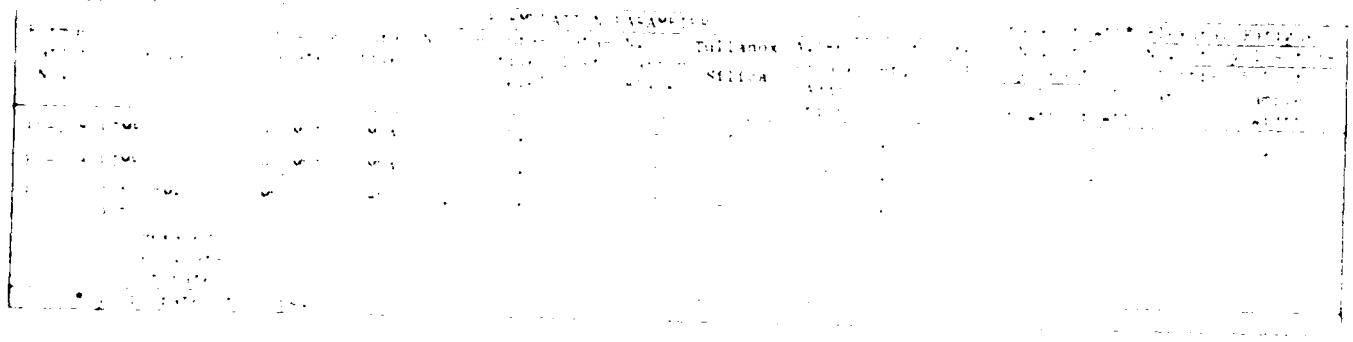
[illegible], are presented in Table 1[illegible]. The materials are prepared by [illegible] knitting, which features a lock stitch that prevents raveling. The polyester [illegible] was selected for use in this program since it possesses strength approaching [illegible] in both warp and fill directions but at a fabric weight of only 14.[illegible] [illegible] oz/yd^2 or 47[illegible] g/m^2. The manufacturer advised that the [illegible] polyester is as good as, or better than, that of the [illegible].

[illegible] polyester knit material were coated with formulation [illegible] of the unreinforced formulation was also used. [illegible] material that the unexpected cure properties of [illegible] were the result of the knitted polyester reinforcement. [illegible] three lower tensile strength than did the fabric [illegible]. [illegible], Appendix B.

TABLE 1. CHARACTERISTICS OF KNIT POLYESTER FABRIC

Material	Source	Weight oz/yd^2	Tensile Strength, pli		Wall Burst, lb
			Warp	Fill	
[illegible]	[illegible]	[illegible]	156	3[illegible]	[illegible]
[illegible]	[illegible]	14.[illegible]	209	35[illegible]	5500

[illegible] knit reinforced PBAN polymer was tested at three temperatures. [illegible] strength, tear strength and coating adhesion results are presented [illegible], at 75°F ambient. These results are presented in Table 19. Since the reinforcing [illegible] tensile and tear strength of reinforced polymers, the [illegible] compared to the results obtained on woven fabric reinforced PBAN [illegible]. A comparison of the results shows that the knit reinforced PBAN has [illegible] the tear strength of the woven fabric reinforced, the same reinforcement weight. However, the [illegible] is approximately three times greater than the woven fabric reinforced material. While the coating adhesion value for the knit material appears [illegible], the type of the polyester knit [illegible] large areas of strike through at the rear [illegible] this case is measuring adhesive strength of the [illegible] rather than the force required to peel the elastomer from the knit. A knit reinforcement structure does not yield as efficient a reinforcement as a woven fabric on a strength-to-weight basis.



*Mr. Paul Lowder, MERADCOM, Fort Belvoir, VA.

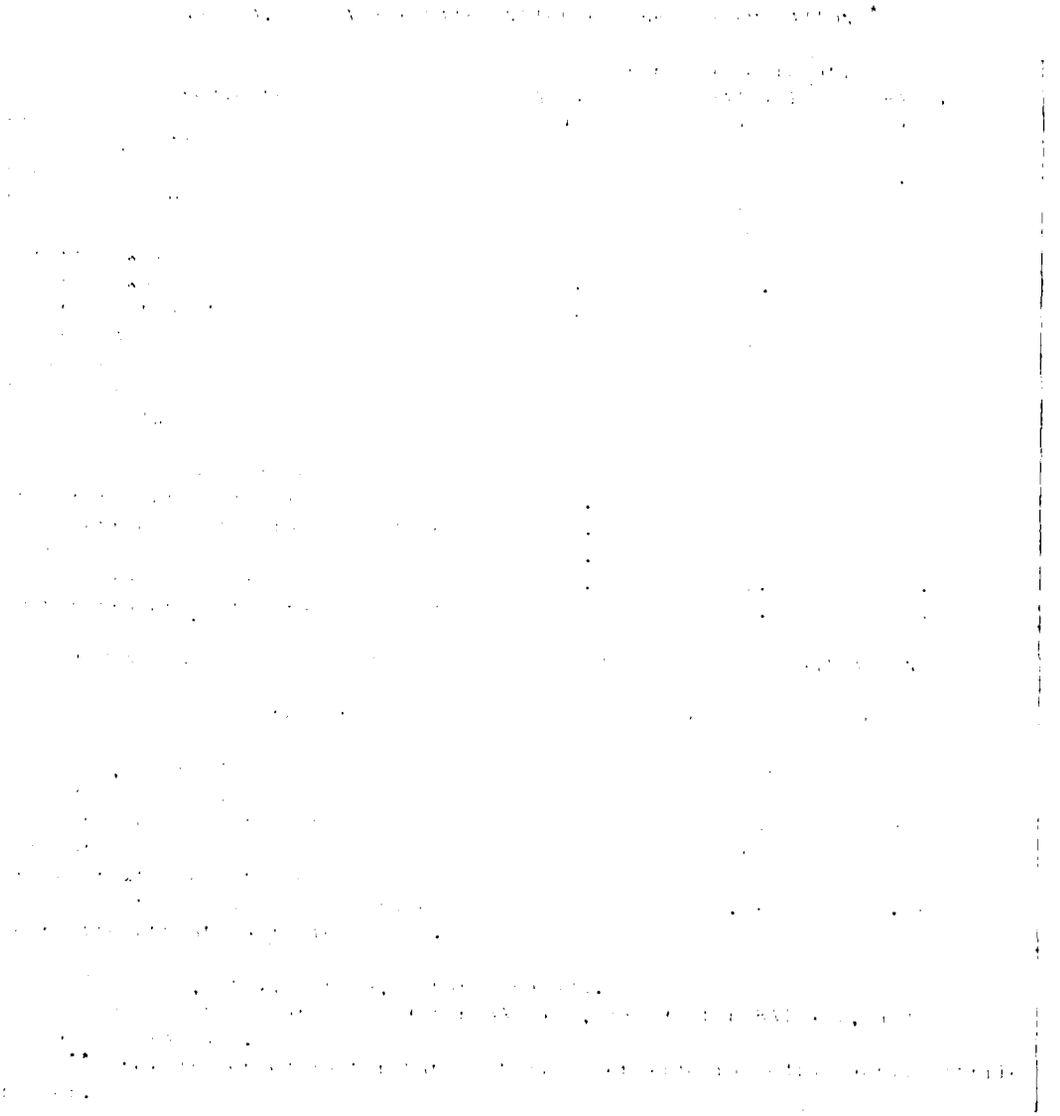
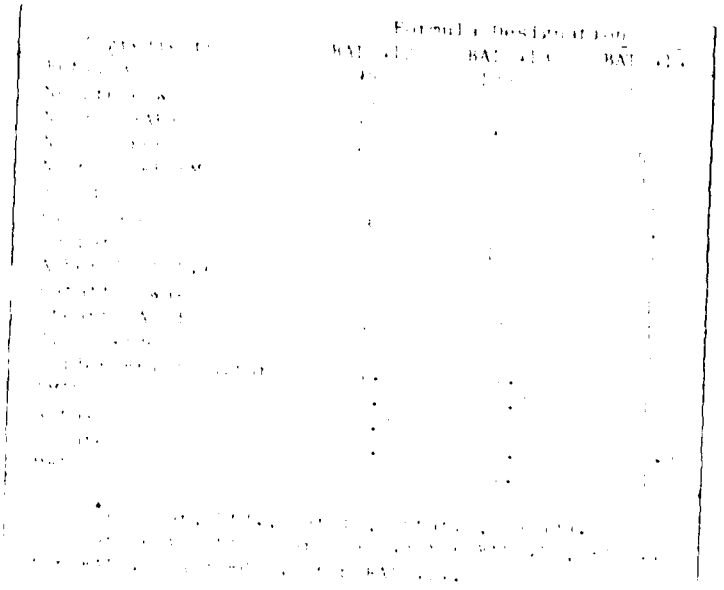
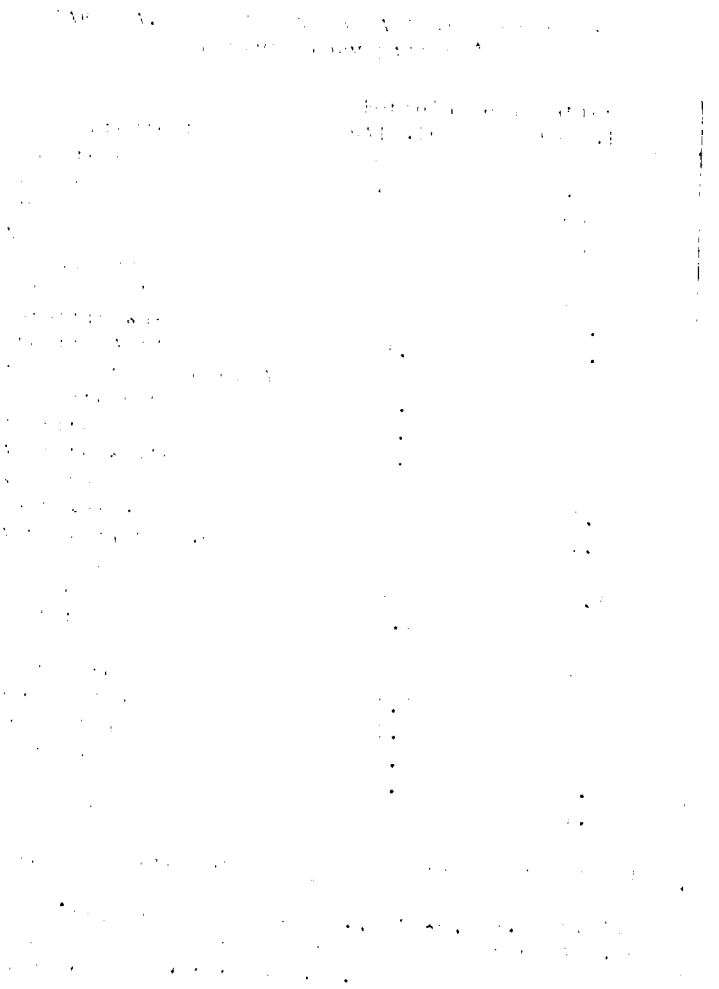


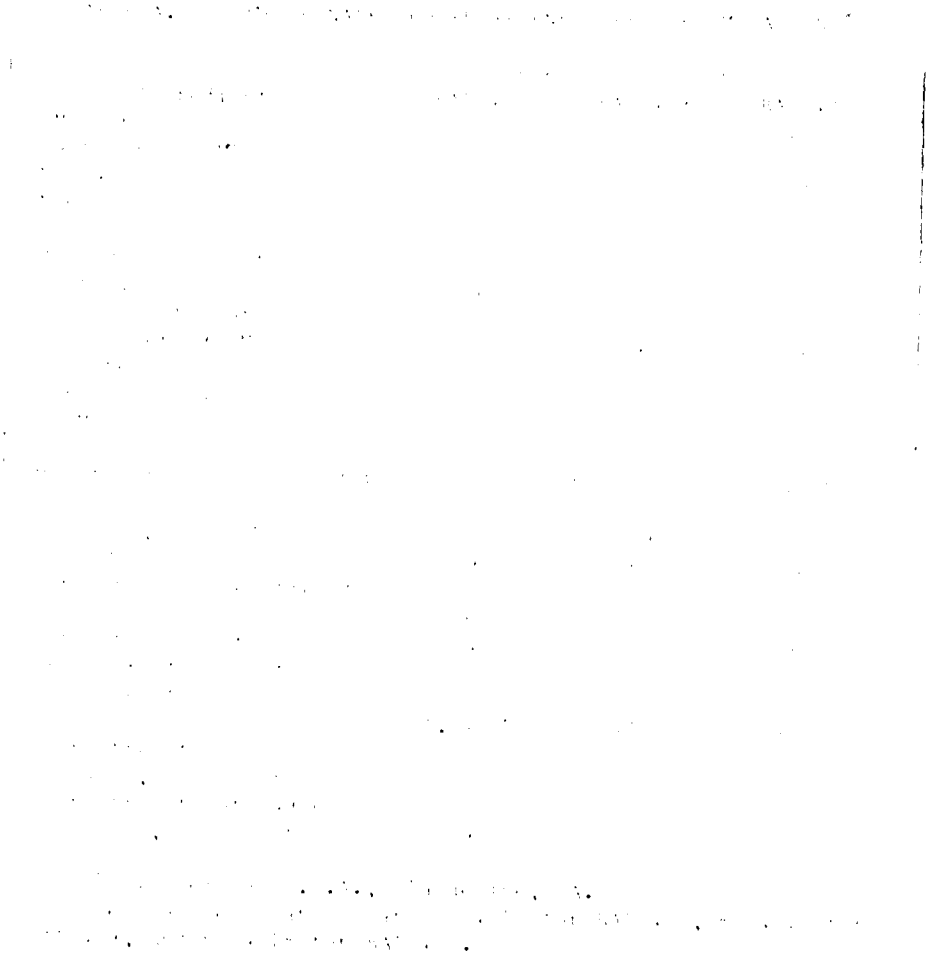
TABLE A.6 — BROMOBUTYL RUBBER FORMULATIONS*

Ingredients	Formula Designation		
	BAT [illegible]	BAT [illegible]	BAT 405
Bromobutyl [illegible]	[illegible]	[illegible]	100
Escalene [illegible]	[illegible]	[illegible]	–
SRF [illegible]	[illegible]	[illegible]	–
[illegible]	[illegible]	[illegible]	–
[illegible]	[illegible]	[illegible]	[illegible]
[illegible]	[illegible]	[illegible]	–
[illegible]	[illegible]	[illegible]	[illegible]
[illegible]	[illegible]	[illegible]	[illegible]
[illegible]	[illegible]	[illegible]	[illegible]
[illegible]	[illegible]	[illegible]	[illegible]
[illegible]	[illegible]	[illegible]	[illegible]
[illegible]	[illegible]	[illegible]	[illegible]
[illegible]	[illegible]	[illegible]	[illegible]
[illegible]	[illegible]	[illegible]	[illegible]
[illegible]	[illegible]	[illegible]	[illegible]

*[illegible] BAT [illegible], SHC-85 for BAT [illegible]







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Calculation of Curative weight

The following equations may be used to calculate the curative addition required:

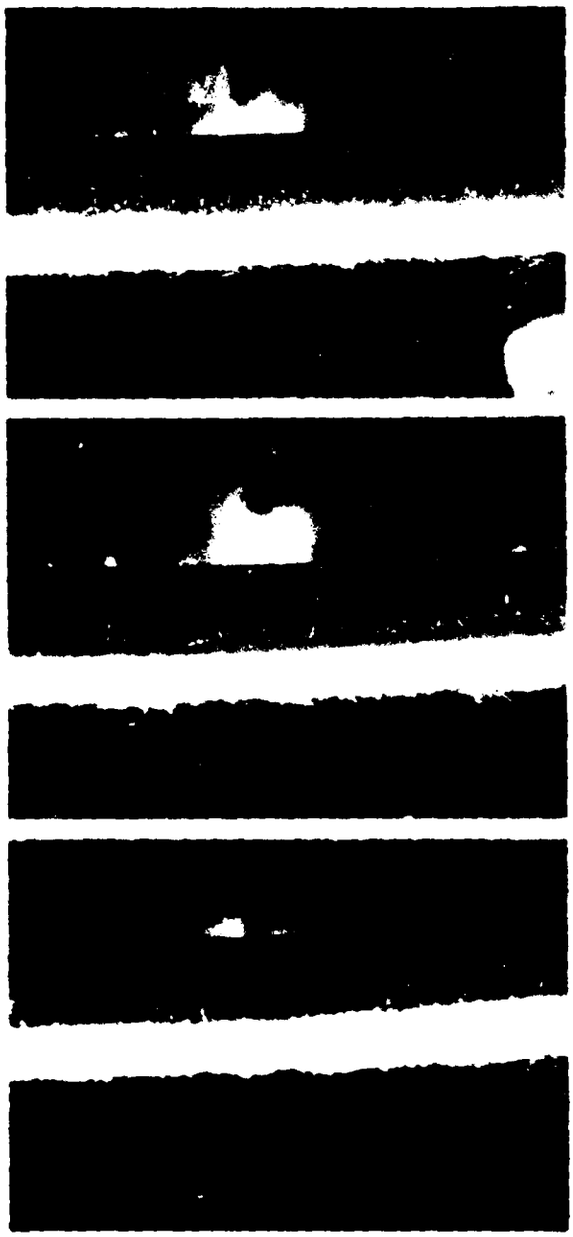
CL = curative level per 100 parts prepolymer

$$\text{Per cent NCO} \times \text{Equivalent Wt. Curative} \times \text{Curative Mole Ratio}$$

$$\text{Curative Mole Ratio} = \frac{\text{Theoretical Curative Level}}{100}$$

TABLE B.1 - TENSILE STRENGTH OF
NYLON AND POLYESTER MAT

Sample No.	Material	Weight per sq. ft.	Tensile strength, lbs. per in.
[illegible]	Nylon Mat	[illegible]	[illegible]
[illegible]	Polyester Mat	[illegible]	[illegible]



a) POLYURETHANE FORMULATION 146

COMPLETELY INFILTRATED MAT.
5.5 oz/yd^2

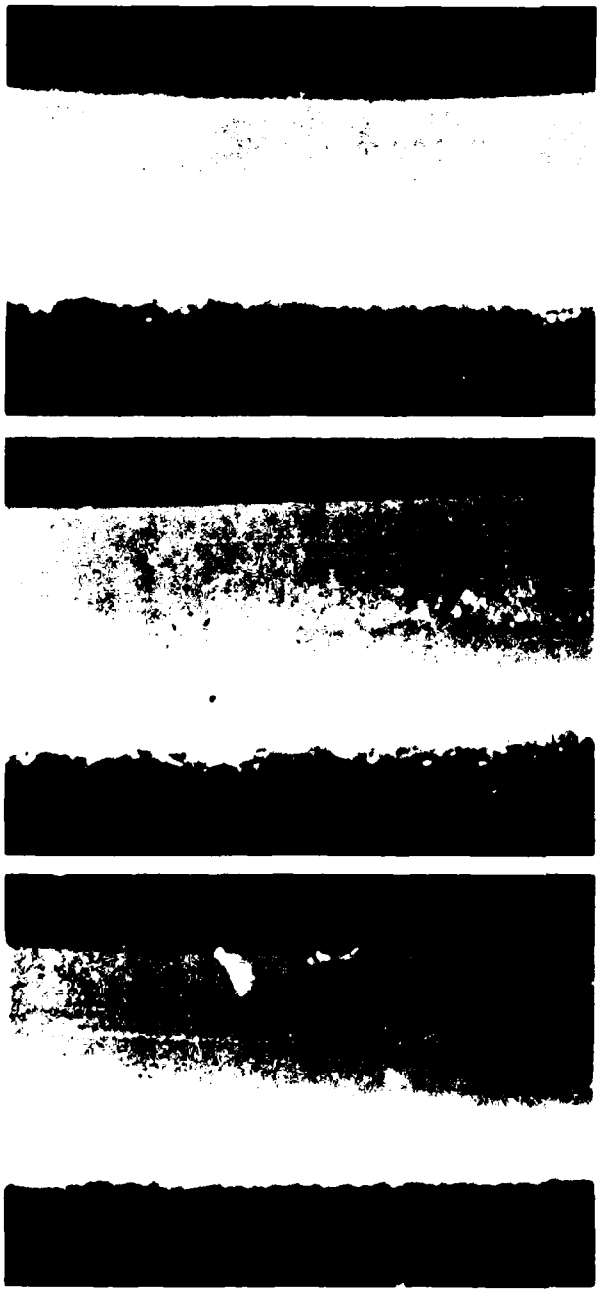
b) POLYURETHANE FORMULATION 146

95-100% INFILTRATED MAT
11.8 oz/yd^2

c) POLYURETHANE FORMULATION 146

INFILTRATED MAT (50-60%)
UNINFILTRATED MAT (40-50%)
14.5 oz/yd^2

Figure B-1 - Permeation of Formulation 205 Polyurethane Into Polyester Mats of Weight 5.5, 11.8 and 14.5 Ounces per Square Yard



a) CAST LIQUID 205 IMPREGNATED MAT. 5.5 oz/yd²

b) CAST LIQUID FORMULATION 205 IMPREGNATED MAT. UNIMPREGNATED MAT. 11.8 oz/yd²

c) CAST LIQUID FORMULATION 205 IMPREGNATED MAT. UNIMPREGNATED MAT. 14.5 oz/yd²

[illegible] ... was noted that ... [illegible]

[illegible] ... ment.

[illegible] NCO ... [illegible]

[illegible] NCO content is defined as the percent by weight of NCO [illegible] ... following [illegible] procedure:

[illegible] ... of the prepolymer, transferred to a dry, [illegible] ... weighed. The sample is then dissolved in [illegible] ml [illegible] ... 0.1N [illegible] ... solution is [illegible] ... to allow [illegible] ... isopropanol and [illegible] drops of [illegible][*] the liquid is titrated with 0.1N HCl solution [illegible] ... similar manner, a "blank" is run using all reagents [illegible].

[illegible] content is then determined using the following calculation:

$$\text{Percent NCO} = \frac{(B - S) \times N \times [\text{illegible}]}{\text{Weight of Sample (g)}}$$

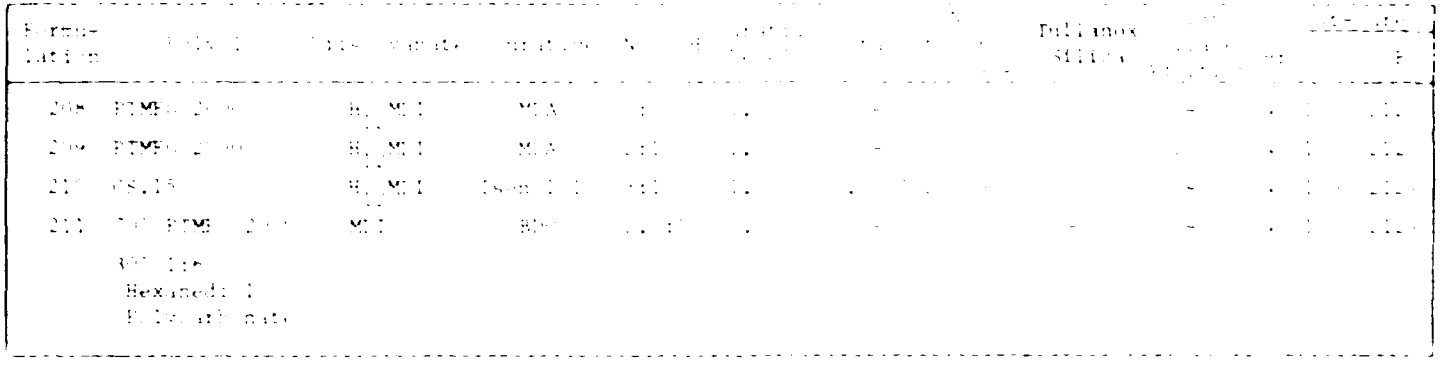
where B = Volume of 0.1N HCl required for blank (ml)

S = Volume of 0.1N HCl required for sample (ml)

N = Normality of HCl solution

---

[*] Bromophenol Blue Indicator: Dissolve 0.1 g dry bromophenol blue in 1.5 ml 0.1N NaOH and dilute to 100 ml with distilled water.



TASK 2 - HOMOGENEOUS MATERIALS DEVELOPMENT

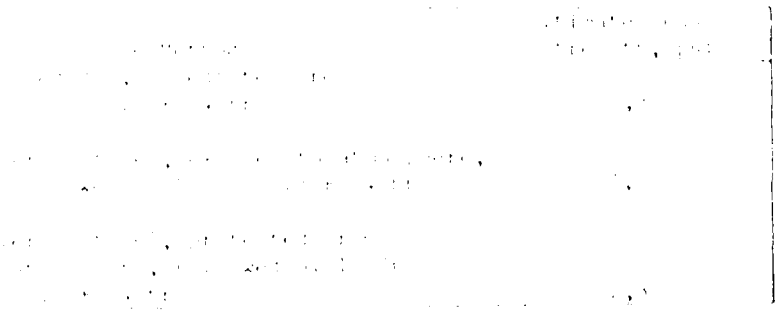
In order to prepare specimens for evaluation in this task, the handling procedures for the polyurethane polymers had to be modified. The addition of 5% carbon black or fumed silica increased the viscosity of the prepolymers to such an extent that the additives could not be added either by simple stirring or in a paint mill.

For formulation PT-270 (BAI-130 + 5% carbon black + 5% fumed silica), approximately 5% of xylene was added to reduce the viscosity, and a satisfactory dispersion was obtained by milling on a paint mill. This material was then stirred with a high- or low-speed stirrer under vacuum to remove the solvent. Analysis of the prepolymer indicated a 2.9% isocyanate content (compared to 3.0% normally expected for this material). The lower NCO figure may have been due to reaction of the isocyanate group with atmospheric moisture, since a large surface area of the prepolymer is exposed during milling. The correct calculated quantity of curative was subsequently stirred into the prepolymer and the material was successfully cast into a 1-yd² sheet.

For PT-211 (PT-270 + 5% carbon black) a slightly different approach was adopted for the modification procedure. The modification was to incorporate the 5% carbon black into the polyurethane prepolymer and, separately, blend the 5% fumed silica into the butane diol chain extender. The two modified components were then to be mixed in a "Semkit" cartridge dispenser, and extruded followed by molding in flat 1 ft x 1 ft sheets.

The carbon black was successfully incorporated into the prepolymer. However, blending the 5% silica into the butane diol was found to be impossible, since only 0.5% part by mass is necessary to satisfy the stoichiometric requirements of the prepolymer (for a prepolymer of 3% NCO content, this would amount to about 5 g of butane diol per 1 kg of prepolymer). Consequently, a compromise was made, and the material was cast containing the carbon black only.

For PT-211 (PT-198 + 5% carbon black and 5% silica) the addition of approximately 5% xylene was necessary, and curing of the polymer was considerably retarded by the presence of the solvent, even though this system was catalyzed with T-12 catalyst. Even post-cure of this material was delayed until most of the solvent had diffused out; otherwise voids would have been formed by the solvent vapor.



[illegible] the material dries the [illegible] when the polymer [illegible] the material allowed to cool while [illegible] that atmospheric moisture was also [illegible] [illegible] polymer [illegible] diffuse into the melted polymer and react with [illegible] to form [illegible].

[illegible] can be obtained only when the material [illegible] immediately [illegible] fires.

[illegible] fabrication work performed [illegible] additional scales were initiated. This was an [illegible] BAI Lot, for specification [illegible] PPMD of m.w. 9[illegible] [illegible] BAI Lot. Other parameters such as %Cl, [illegible] [illegible]. The lot, as at this investigation was [illegible] [illegible] was for not exceeding lower mobility [illegible] [illegible] material of higher molecular [illegible].

[illegible] evaluated in such screening [illegible] of strength of the fabric reinforcement test scales.

[illegible] Water Barriers

[illegible] testing, two polyurethane formulations, [illegible] selected to prepare water barrier samples. [illegible] as described earlier for the elastomer polymer work. [illegible] prepared at a solids of 1 gal. Prior to use, the 1-gal [illegible] were subjected to vacuum to remove air content [illegible] [illegible] The fabric was then treated [illegible] and allowed to dry.

[illegible] fabric was coated with the liquid elastomer using [illegible]. Previous work at BAI demonstrated that, [illegible] the fabric after the application of the liquid elastomer [illegible] the fabric at the filament intersections, despite [illegible] [illegible] After gelling, the partially [illegible] were turned over and coated on the opposite side. [illegible], the coated sheets were post-cured at 100°C (212°F) for 4 hr. [illegible] were prepared for testing at BAI, and for delivery to the Center [illegible]. In addition to the coated fabric specimens, 1-yd² sheets of the [illegible] were also prepared for delivery to the Center.

[illegible] polyurethane formulation 2[illegible], a difference in processing was necessary in the fabrication of large size (1 yd²) sheets of the fabric reinforced material which were required. Whereas the properties of the original formulation were derived from test sheets prepared by heat curing of the liquid elastomer [illegible] flow immediately after casting, the polymers applied to the 1-yd² sheets of fabric were allowed to gel overnight to facilitate handling; they were then oven-cured at 100°C (212°F) for 4 hr.

[redacted]

PBD	R-45M				CSTS			
No.	IPDI		H_{12}MDI		IPDI		H_{12}MDI	
Curative	BDO	MDA	BDO	MDA	BDO	MDA	BDO	MDA
Viscosity	197	191	187	186	19	189	186	18•

In general, all four prepolymers, when treated with MDA as a curative, reacted at room temperature with ambient moisture to yield opaque light brown rubbery materials in a few hours. The short reaction times immediately eliminated these materials from further consideration.

The four prepolymers, when treated with BDO as a curative, reacted very slowly and it was necessary to catalyze the reaction with 0.02 wt% DBTDL at least. These mixtures were poured into molds and were then post-cured at 100°C for 2 hrs. They were clear, tough to low modulus elastomers, but, as will be seen in the tensile data presented in Appendix C.

The work in this case was supplemented as follows: An additional 25% by wt. R-45M, was formulated at a NCO:OH ratio of 3:1 to prepare a series of higher solids materials. However, these materials did not possess adequate flowability for processing at 60°C with reasonable pot lives.

Since the initial work was performed at the Center, it was decided to further evaluate the prepolymers with a different set of curative agents, Isonol 1:3 DEA, N,N,N',N'-tetrakis (2-hydroxypropyl) ethylene diamine (Quadrol) and 2-ethyl 1:3 hexane diol with a potential yield of 200 gms. In place of stannous octoate catalyst heretofore, the prepolymers were treated with dibutyltin dilaurate. The prepolymer R-45M yielded soft, gummy products; those from CSTS were more viscous, but were free-pouring, and led to an additional four elastomers to evaluate. Formulations for prepolymers at an NCO:OH ratio of 3:1 according to the following experimental plan.

	CSTS			
	IPDI		H_{12}MDI	
Curative	2 et. 1:3*	Isonol	2 et. 1:3*	Isonol
Yield, gms	199	200	197	198

*2-ethyl 1:3 hexane diol

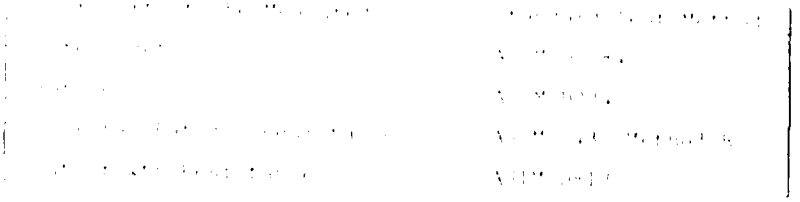


TABLE [illegible] SUMMARY OF EVALUATION [illegible] FOR RUBBER COATED FABRICS

Properties To Be Measured	Standard Test Method
[illegible]	ASTM [illegible]
[illegible]	[illegible]
[illegible] Adhesion	ASTM [illegible]
[illegible]	ASTM [illegible]
[illegible] Tear Strength of Textile Fabrics	ASTM [illegible]
Water Vapor Transmission	Weight Gain

A Scotchcal [illegible] was applied to the rubber.

APPENDIX D

DETAILED EXPERIMENTAL DATA

This appendix presents more detailed experimental data from which the averaged results were calculated and presented in the text. In most cases, the results for each test specimen are tabulated; in a few cases where the numerical spread in data was small, the maximum and minimum values are tabulated in addition to the average value, rather than each individual data point. Usually 3 to 5 data points were included in each average value.

In Tables D.1 and D.2 are presented the crack growth and flexural fatigue data for the gum rubber polymers and the castable liquid polyurethane polymers, respectively. Results are presented for each of the three specimens which were evaluated for each polymer, and the generally large spread in results of this type of test is obvious.

Tables D.3 and D.4 contain the mechanical characterization data for the gum rubber polymers and the castable liquid polyurethane polymers which were selected for this phase of the program. Table D.5 presents the coating adhesion data. Tables D.6 and D.7 contain the Pico abrasion test results for the gum rubber polymers and the castable liquid polyurethane polymers. Experimental results for the duplicate specimens which were evaluated with the Goodrich flexometer hysteresis test are listed in Tables D.8 and D.9 for the gum rubbers and polyurethanes, respectively. Plots of the temperature rise versus time for the various polymers are presented in Figures D.1, D.2, D.3, D.4, D.5, D.6, and D.7 for the gum rubbers and in Figures D.8, D.9, and D.10 for the polyurethane polymers.

More details of the experimental data characterizing the effects of water on the polymers are presented in Tables D.10 and D.11. Similarly, the effects of heat conversion are presented in Tables D.12 and D.13.

Tensile and elongation data for each specimen at three different test temperatures are presented in Table D.14 for the gum rubber polymer coated nylon fabric laminate and for the castable polyurethane polymer coated nylon fabric laminate in Table D.15. Similarly, tear strength results are presented in Tables D.16 and D.17. Coating adhesion results are presented in Table D.18 for polyurethane coated nylon fabric; the fabric and adhesion promoter were the same as used in the previous coating adhesion tests.

[illegible] for the tri-ply dry specimens of polymer coated nylon fabric laminates [illegible] evaluated for flexural fatigue resistance in the DeMattia test are [illegible] Tables 9.19 and 9. .

[illegible] water vapor and oxygen transmission results are presented as a function of [illegible] Tables 9.20 and 9.21.

[illegible] that the wet fabric composite specimens are presented [illegible] Tensile strength, 9.2) Tear strength, and 9.23 Coating adhesion.

Polygon Designation	Number of Grids					
BAT-146	***	***	***	***	***	***
PL-183	***	***	***			
PL-185	***	***	***			
PL-187	***	***	***	***	***	***
PL-188	***	***				
PL-192						
PL-193						
PL-196	***	***	***			
PL-198	***	***	***			
PL-198.2	***	***	***			
PL-204	***	***	***	***	***	***
PL-205						
PL-206	***	***	***	***	***	***

*

**

toward the thickness

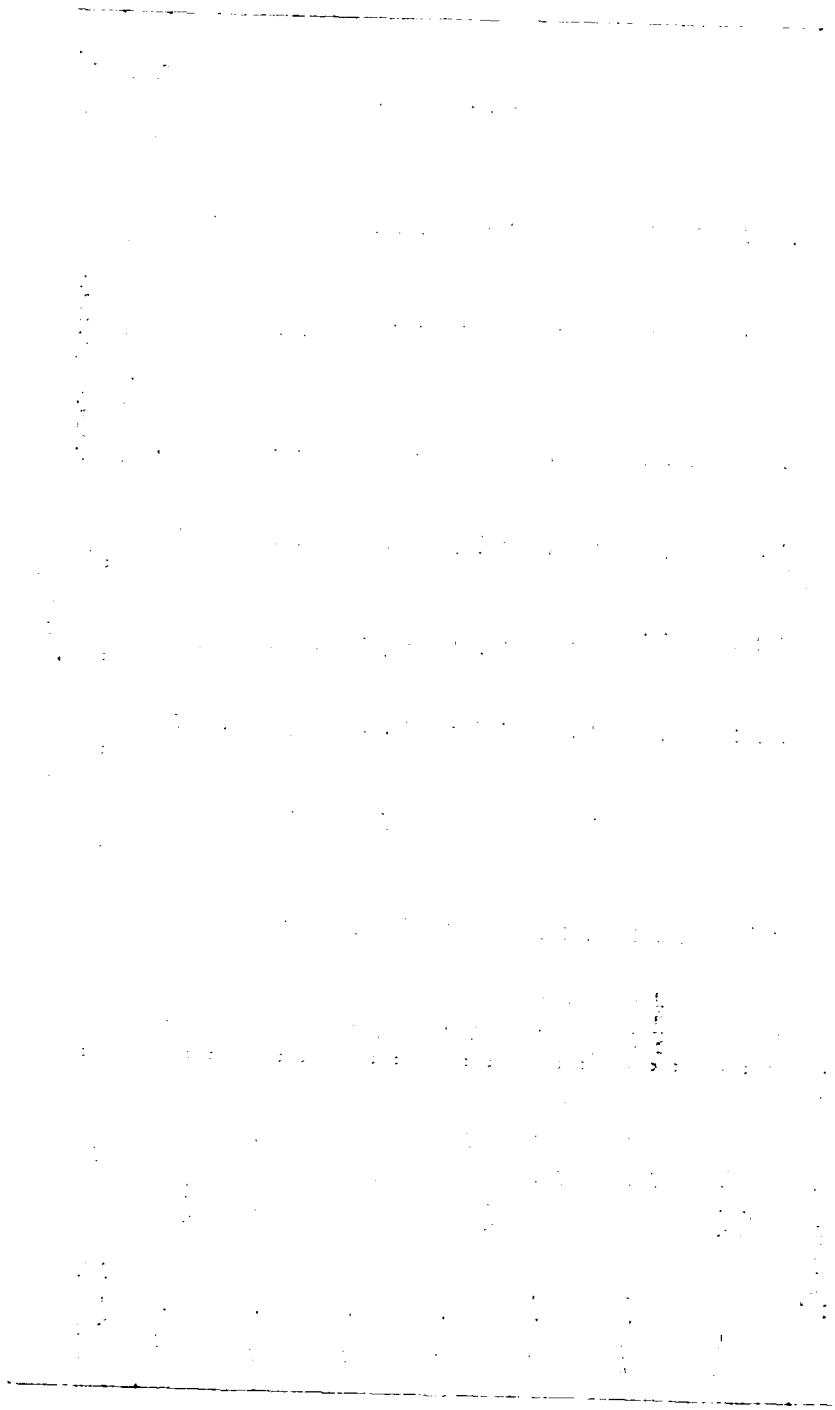
thickness, the

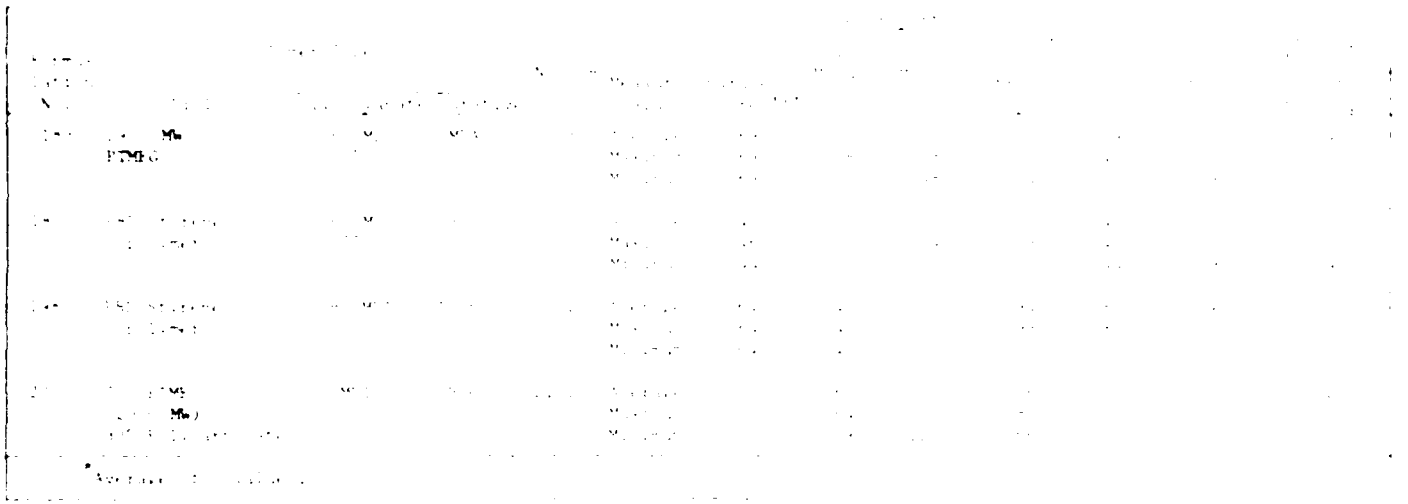
***Sample thickness

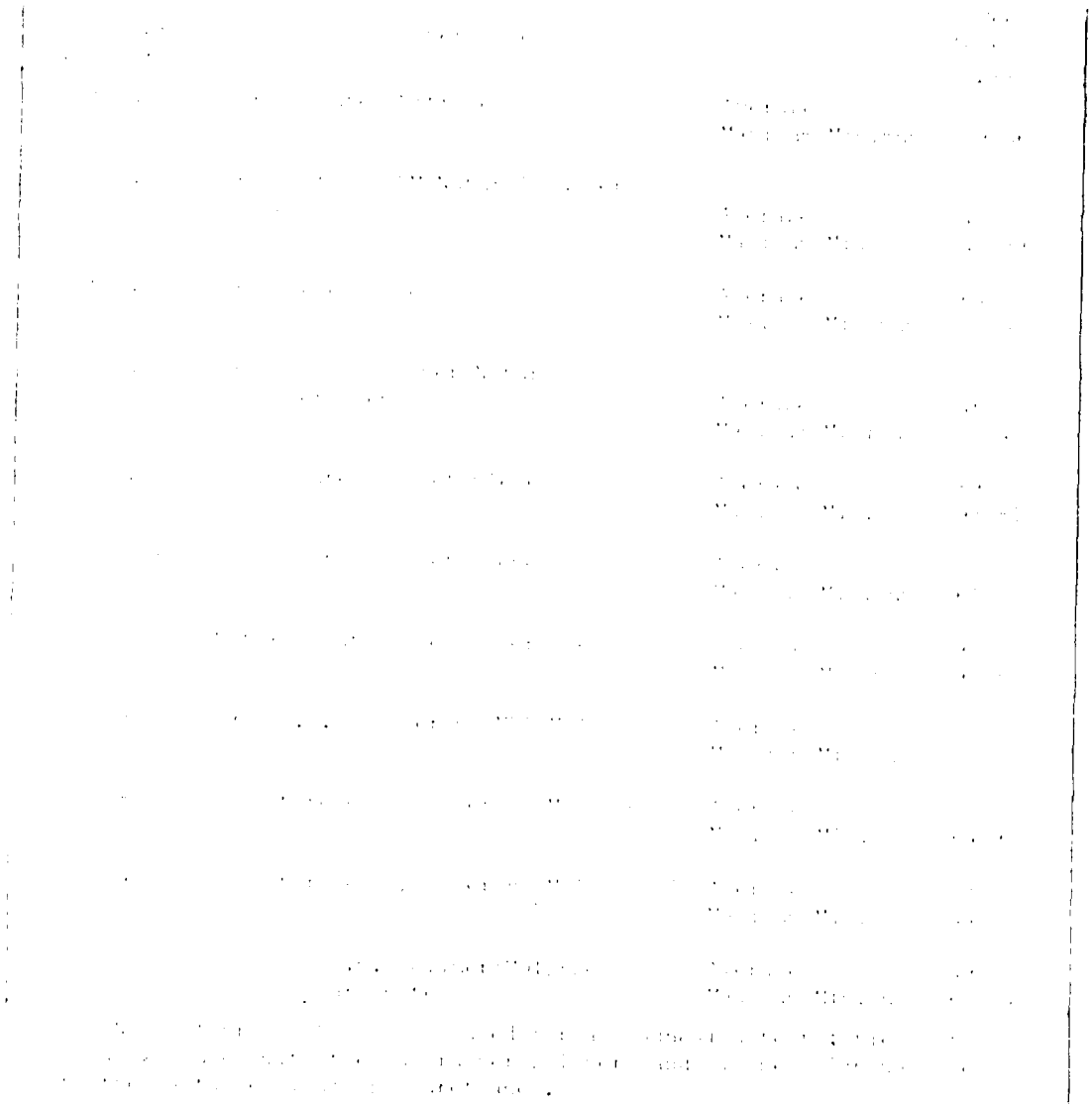
Results were

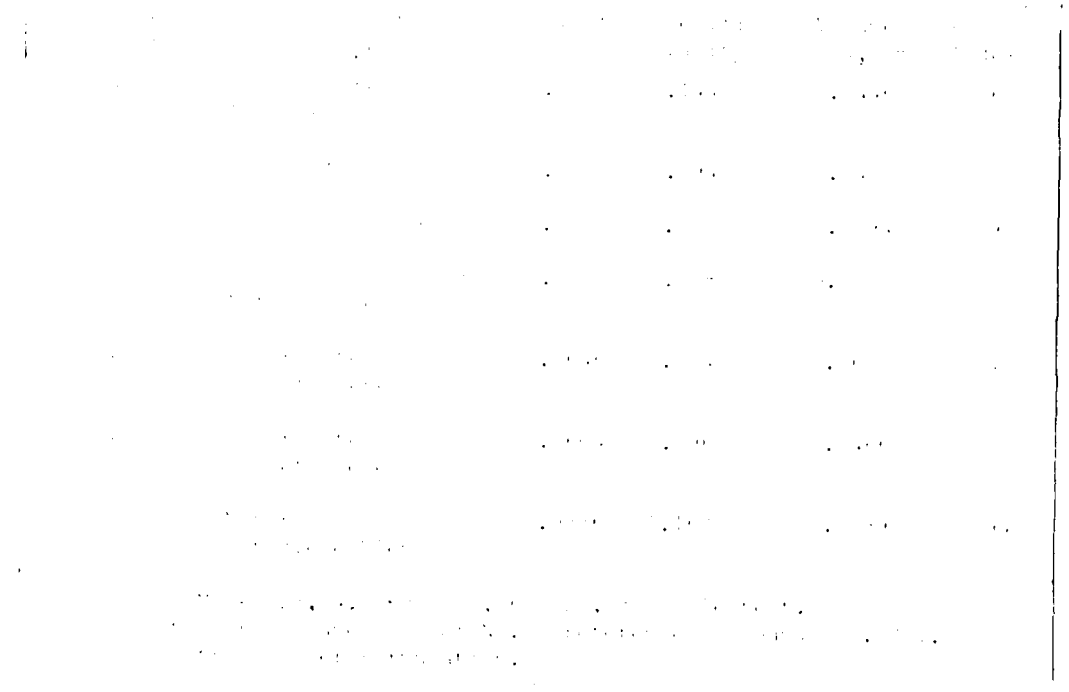
*Full width

**One-half width









AD A151 438 FINGER MATERIALS FOR AIR CUSHION VEHICLES VOLUME 1 FLEXIBLE COATINGS FOR..(U) BELL AEROSPACE TEXTRON BUFFALO NY P K CONN ET AL. DEC 84 7467-927048-VOL-1 2/2

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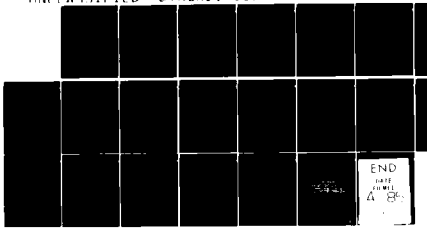
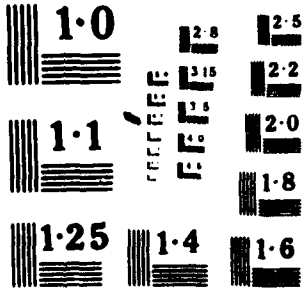



1·0
1·1
1·25
1·4
1·6
1·8
2·0
2·2
2·5
2·8
3·15
3·6

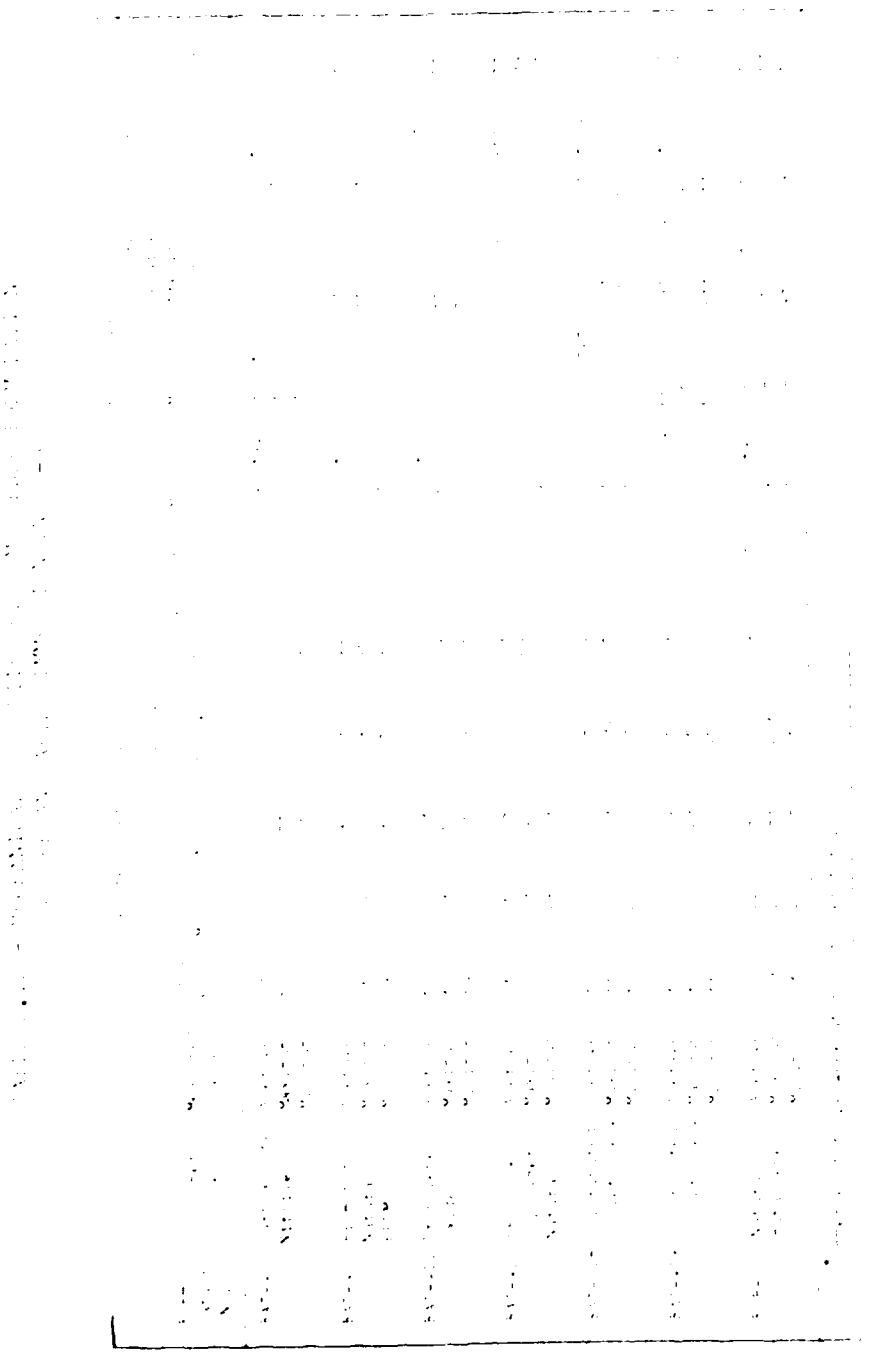


TABLE 3.13 - MECHANICAL PROPERTIES OF POLYURETHANE FORMULATIONS BEFORE AND AFTER IMMERSION IN [illegible]

Formulation No.	Polyol	Isocyanate	Catalyst	Before Immersion						After Immersion in [illegible]					
183	2000 MW PTMEG														
185	PBD Styrene Copolymer														
198	PBD Styrene Copolymer														
205	70% PTMEG (2000 MW) 30% 1,4-Butanediol														

[illegible] immersed 1 [illegible] one week at [illegible] temperature.

TABLE [illegible] - TENSILE RESULTS FROM RUBBER/WOVEN NYLON COMPOSITE

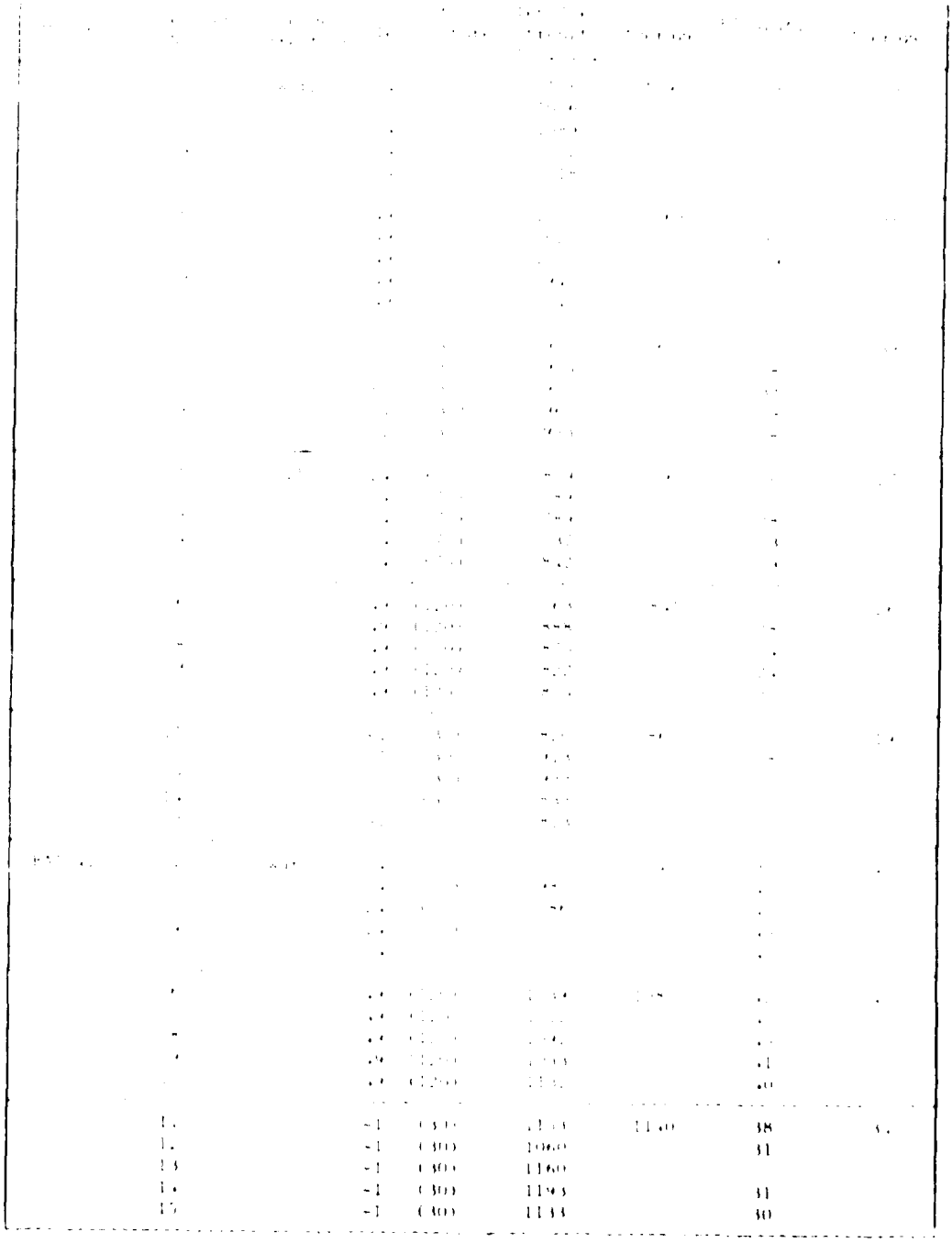


TABLE D.14 - (Continued)

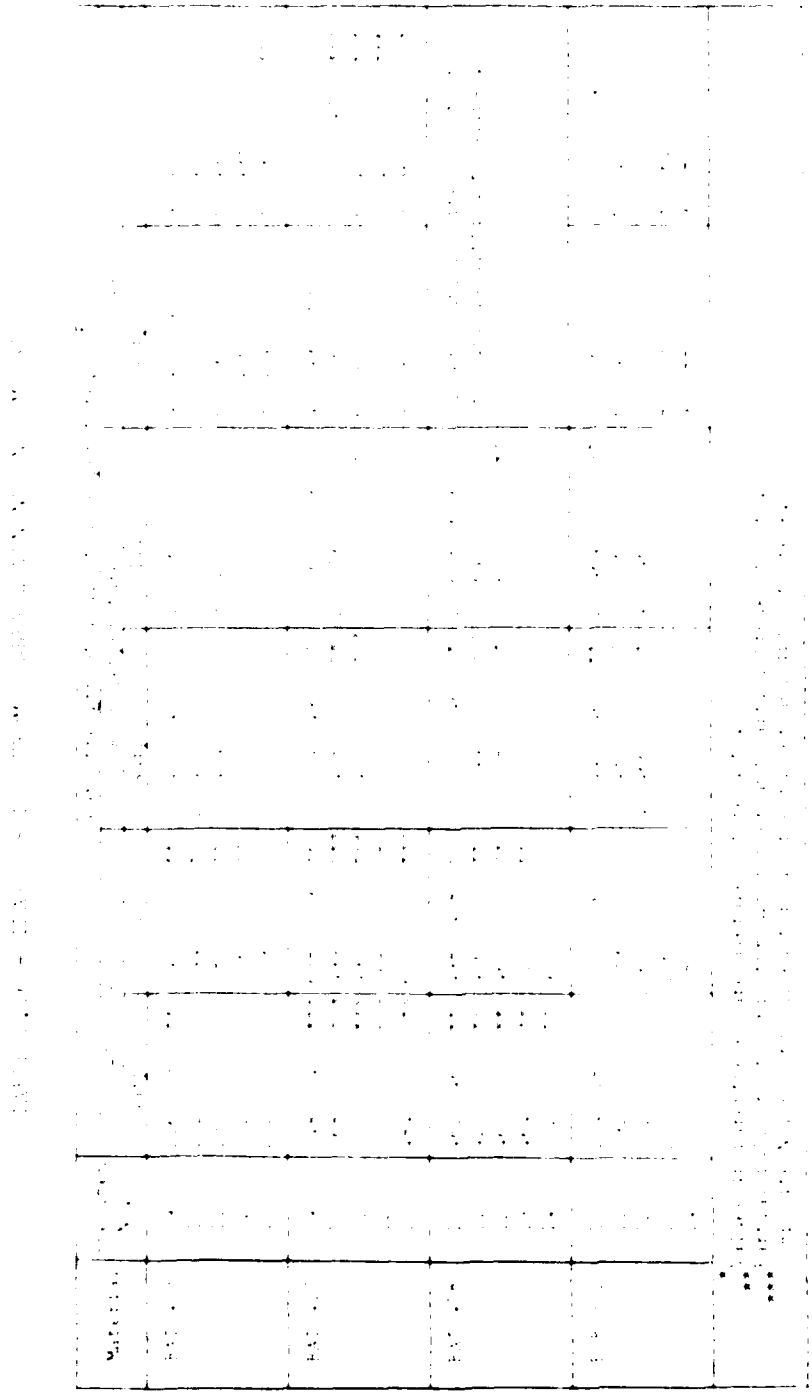
				Test Temperature °F	Tensile Strength lbs./in.	Average	Elongation %	Average
					1020	1052	33	26
					1086		24	
					993		25	
					1052		31	
					1052		30	
					1040	1067	25	29
					1112		30	
					1062		30	
					1078		28	
					1096		31	
					1080	1064	-	24
					1086		24	
					1044		24	
					1060		-	
					1042		20	
					1033	1064	30	24
					994		31	
					987		25	
					1006		25	
					1033		24	
					1020	1054	27	31
					1039		38	
					1086		34	
					1092		34	
					1070		37	
					1108	1064	-	32
					994		28	
					894		42	
					987		34	
					981		34	
					987	1029	14	17
					1086		19	
					987		17	
					1033		14	
					1033		20	
					1003	1033	28	27
					1006		26	
					1070		26	
					1099		30	
					1046		30	

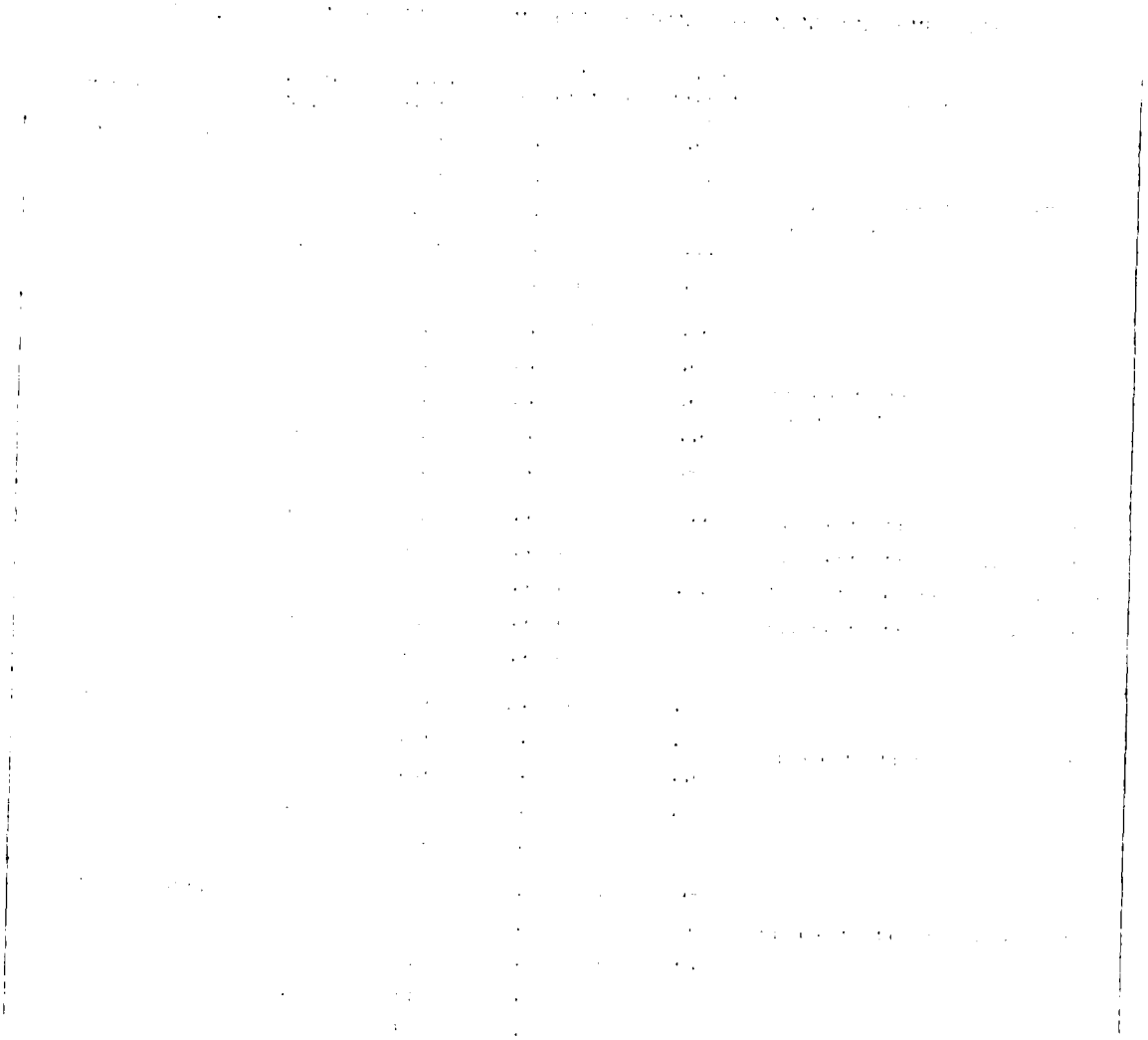
TABLE D.14 - (Continued)

Material	Specimen No.	Weave Direction	Test Temperature °F (°C)	Tensile Strength lbs. in.	Average	Elongation %	Average
	11		-1 (-80)	1492	1068	35	25
	12		-1 (-80)	1642			
	13		-1 (-80)	1258		20	
	14		-1 (-80)	972		17	
	15		-1 (-80)	1041		14	
BAC 418	1	Warp	74 (23)	1327	1,90	44	44
	2		74 (23)	2014		49	
	3		74 (23)	1173		44	
	4		74 (23)	1151		46	
	5		74 (23)	1151		46	
	6		49 (1,200)				
	7		49 (1,200)				
	8		49 (1,200)	Testing was cancelled,			
	9		49 (1,200)	material returned.			
	10		49 (1,200)				
	11		-1 (-80)				
	12		-1 (-80)				
	13		-1 (-80)	Testing was cancelled.			
	14		-1 (-80)				
	15		-1 (-80)				
	1	Fill	74 (23)	1119	1,91	30	30
	2		74 (23)	1224		30	
	3		74 (23)	1125		30	
	4		74 (23)	1220		29	
	5		74 (23)	1306		29	
	6		49 (1,200)	1152	1129	25	28
				Rubber covering gave-way first pull; spec. repositioned and test completed.			
	7		49 (1,200)	1289		32	
				(Test temperature = 1.3°F) first test at 129°F; rubber gave-way.			
	8		49 (1,200)	Rubber covering softened and was separated from fabric by grip.			
	9		49 (1,200)	1089		26	
				(Test Temperature = 110°F)			
	10		49 (1,200)				
	11		-1 (-80)	Testing cancelled.			
	12		-1 (-80)				
	13						
	14						
	15						

TABLE 5.1-1 - TENSILE RESULTS FROM ACCELERATED WATER
NYLON COMPOSITE

Material	Specimen No.	Weave Direction	Test Temperature °F	Test Temperature °C	Material Thickness in.	Tensile Strength lbs./in.	Elongation
Nylon Fabric	1	Warp	24	(75)	0.142	1040	35
	2	Warp	24	(75)	0.184	1100	45
	3	Warp	24	(75)	0.200	1160	80
						Avg. 1100	Avg. 67
	4	Warp	69	(120)	0.172	1150	50
	5	Warp	69	(120)	0.198	1030	40
	6	Warp	69	(120)	0.195	1100	35
	7	Warp	69	(120)	0.207	1040	35
						Avg. 1080	Avg. 37
Nylon Fabric	1	Fill	24	(75)	0.184	830	20
	2	Fill	24	(75)	0.186	1180	25
	3	Fill	24	(75)	0.185	1170	30
	4	Fill	24	(75)	0.191	1290	25
						Avg. 1120	Avg. 25
		Fill	69	(120)	0.168	1120	30
	6	Fill	69	(120)	0.172	970	8
	7	Fill	69	(120)	0.186	1120	18
	8	Fill	69	(120)	0.187	1000	22
						Avg. 1070	Avg. 19
181 - Nylon Fabric	1	Warp	24	(75)	0.140	1140	5
	2	Warp	24	(75)	0.139	1180	30
	3	Warp	24	(75)	0.139	1060	30
	4	Warp	24	(75)	0.140	1240	35
						Avg. 1130	Avg. 25
	5	Warp	69	(120)	0.140	1060	32
	6	Warp	69	(120)	0.144	1040	40
	7	Warp	69	(120)	0.140	1100	35
	8	Warp	69	(120)	0.146	1080	32
						Avg. 1080	Avg. 34
181 - Nylon Fabric	1	Fill	24	(75)	0.1[illegible]	1100	25
	2	Fill	24	(75)	0.170	1100	22
	3	Fill	24	(75)	0.169	1120	26
	4	Fill	24	(75)	0.170	1220	25
						Avg. 1160	Avg. 25
	5	Fill	69	(120)	0.151	960	25
	6	Fill	69	(120)	0.150	870	23
	7	Fill	69	(120)	0.152	970	28
	8	Fill	69	(120)	0.150	1030	22
						Avg. 960	Avg. 25



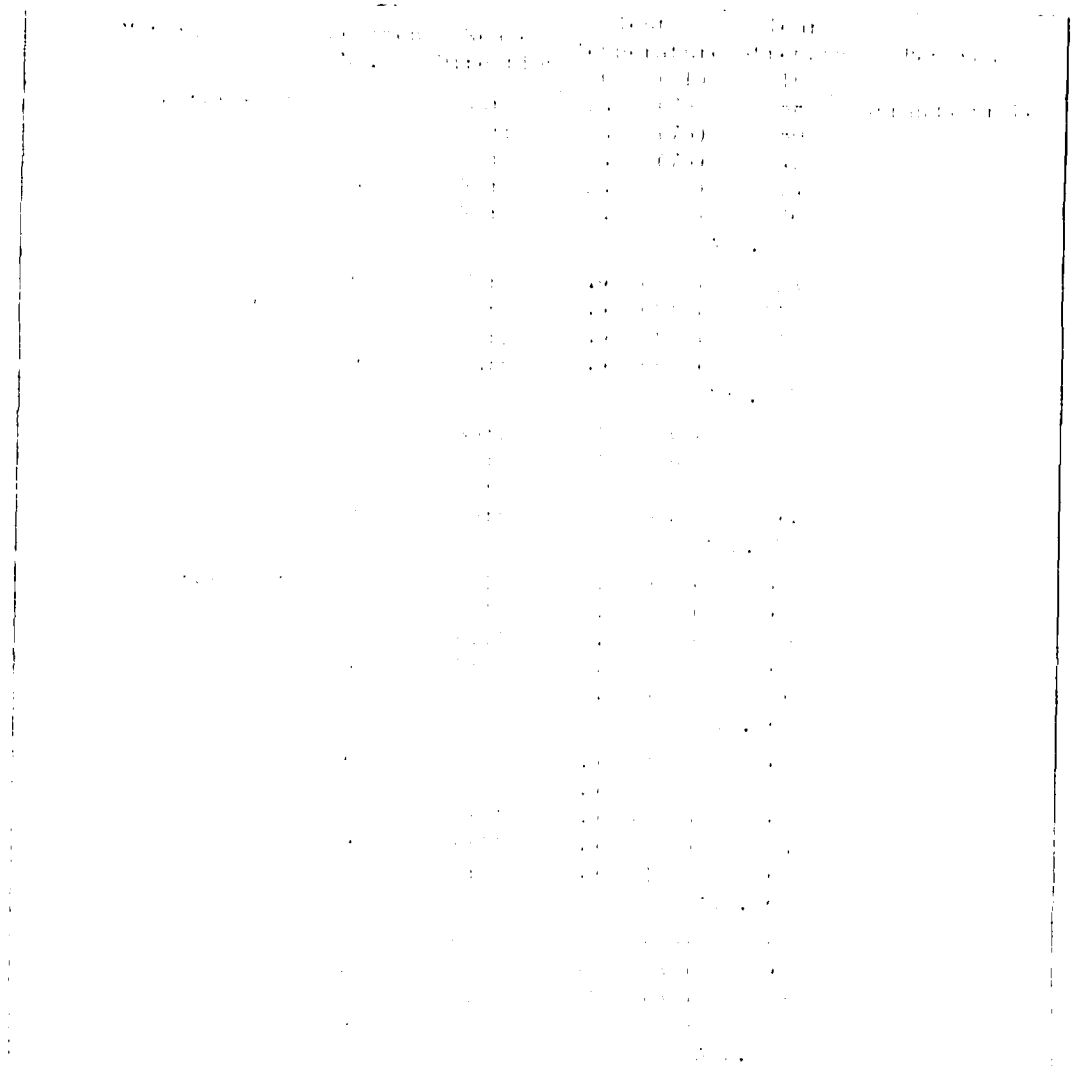


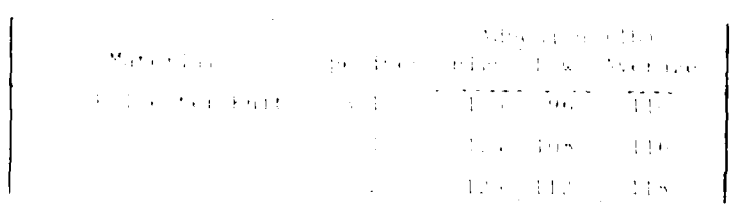
Material Description	Test Temperature °C	(°F)	Crack Initiation	[illegible]	[illegible]*	[illegible]
FCB 402 Nylon Warp	[illegible]	[illegible]	[illegible]	[illegible]	[illegible]	[illegible]
	49	(120)	[illegible]	[illegible]	[illegible]	[illegible]
Fill	[illegible]	[illegible]	[illegible]	[illegible]	[illegible]**	[illegible]**
	49	(120)	[illegible]	[illegible]	[illegible]**	[illegible]**
BAI 403 Nylon Warp	[illegible]	[illegible]	[illegible]	[illegible]	[illegible]	[illegible]
	49	(120)	[illegible]	[illegible]	[illegible]	[illegible]
Fill	[illegible]	[illegible]	[illegible]	[illegible]	[illegible]	[illegible]
	49	(120)	[illegible]	[illegible]	[illegible]	[illegible]
BAI [illegible] Nylon Warp	[illegible]	[illegible]	[illegible]	[illegible]	[illegible]	[illegible]
	49	(120)	[illegible]	[illegible]	[illegible]	[illegible]
Fill	[illegible]	[illegible]	[illegible]	[illegible]	[illegible]	[illegible]
	49	(120)	[illegible]	[illegible]	[illegible]	[illegible]
BAI [illegible] Nylon Warp	[illegible]	[illegible]	[illegible]	[illegible]	[illegible]	[illegible]
	49	[illegible]	[illegible]	[illegible]	[illegible]	[illegible]
Fill	[illegible]	[illegible]	[illegible]	[illegible]	[illegible]	[illegible]
	49	[illegible]	[illegible]***	[illegible]***	[illegible]***	[illegible]***

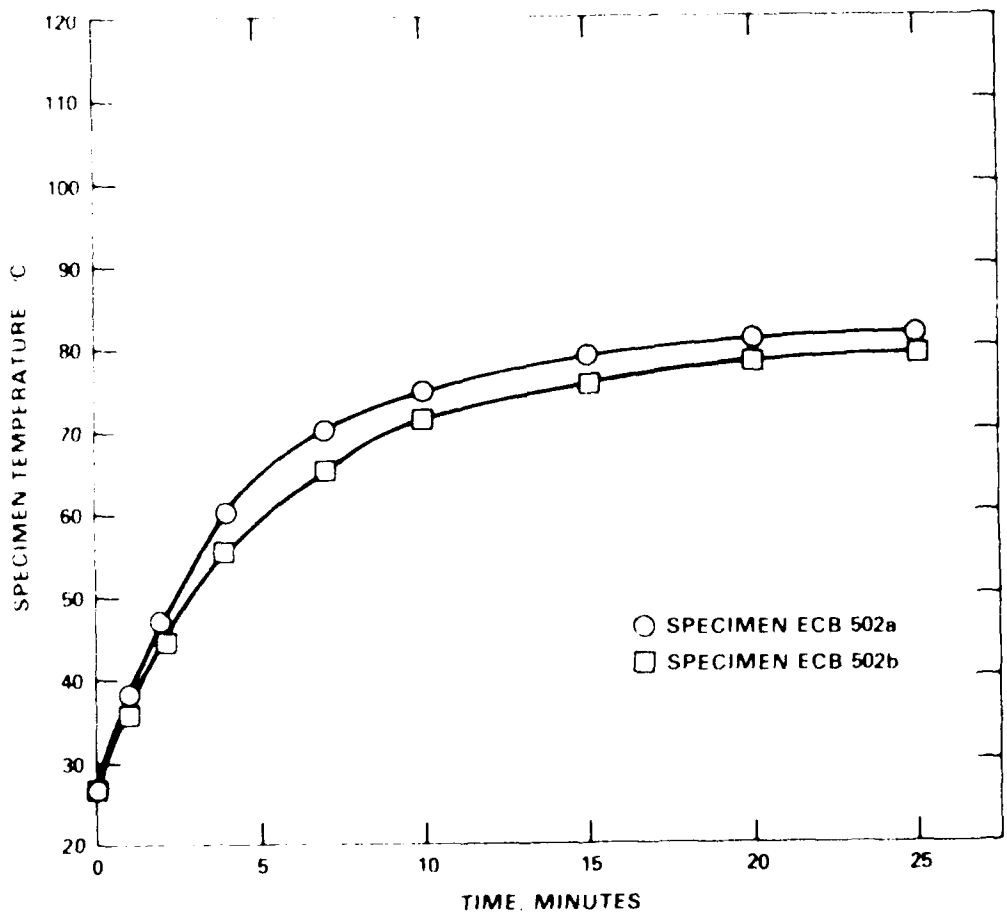
* The data [illegible]

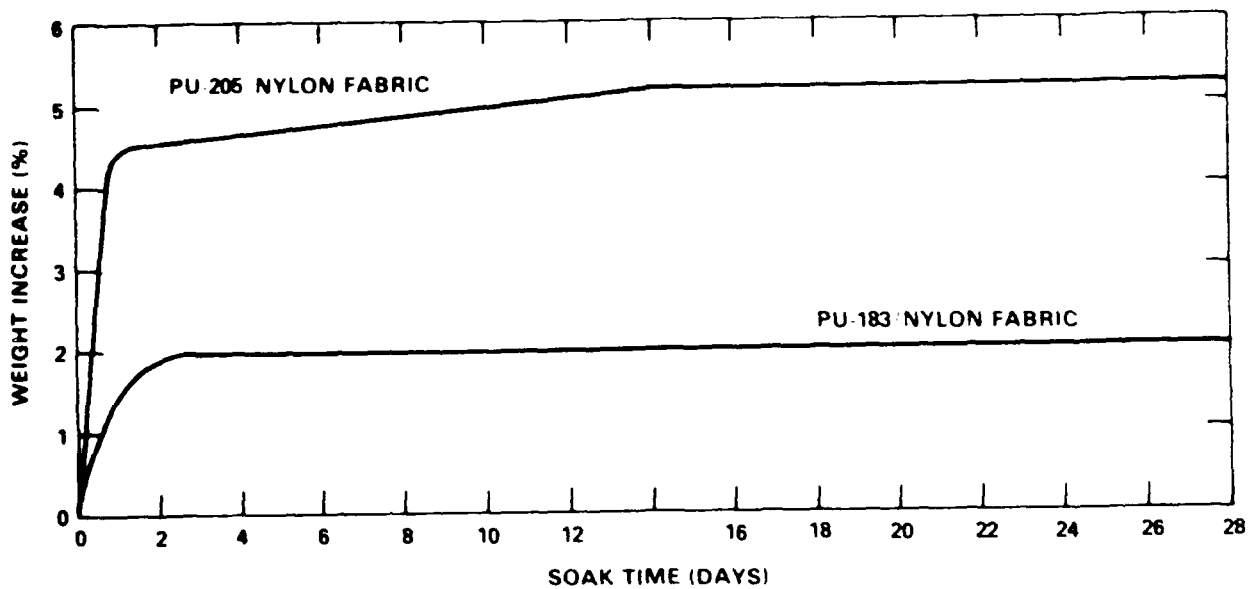
** Specimens exhibited [illegible]

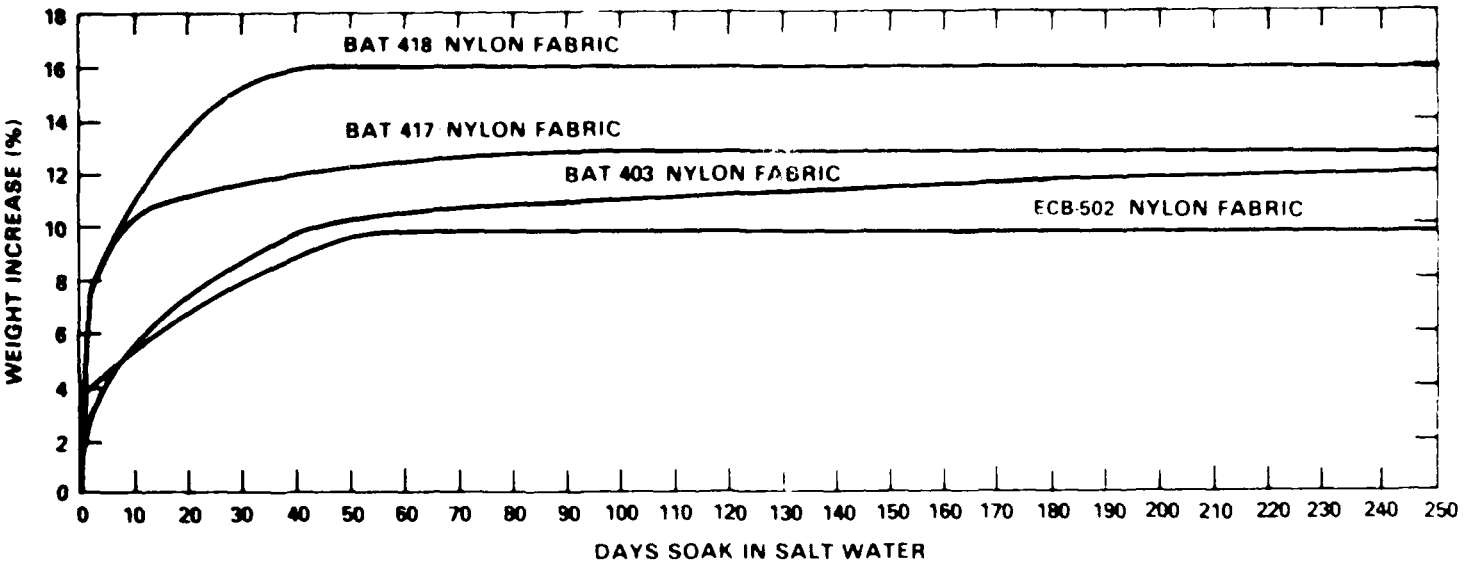
*** Specimens exhibited [illegible]

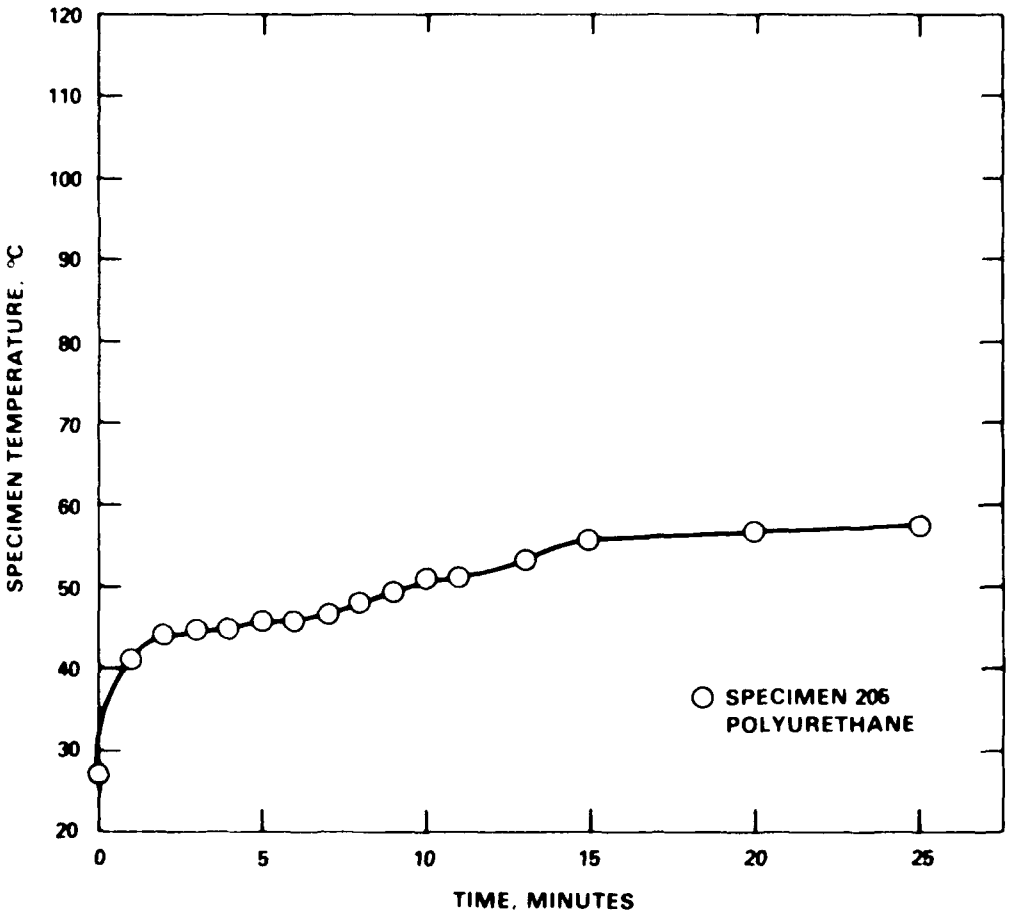


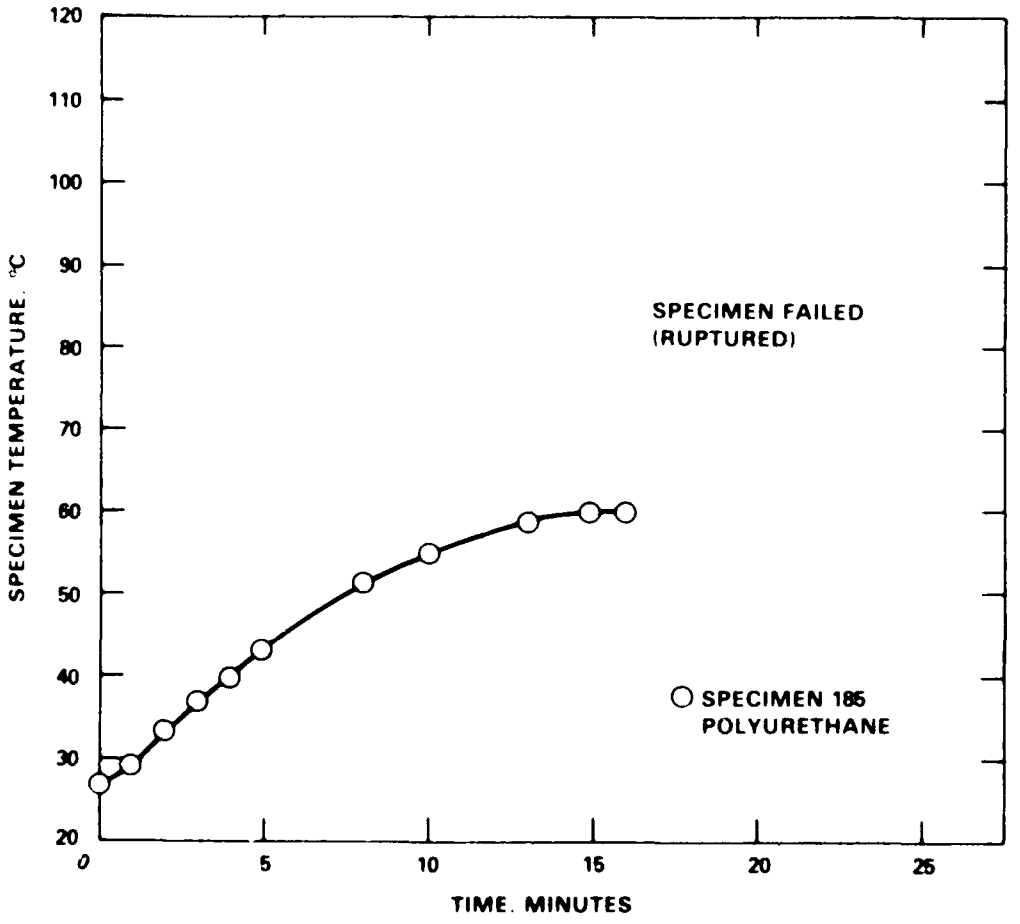


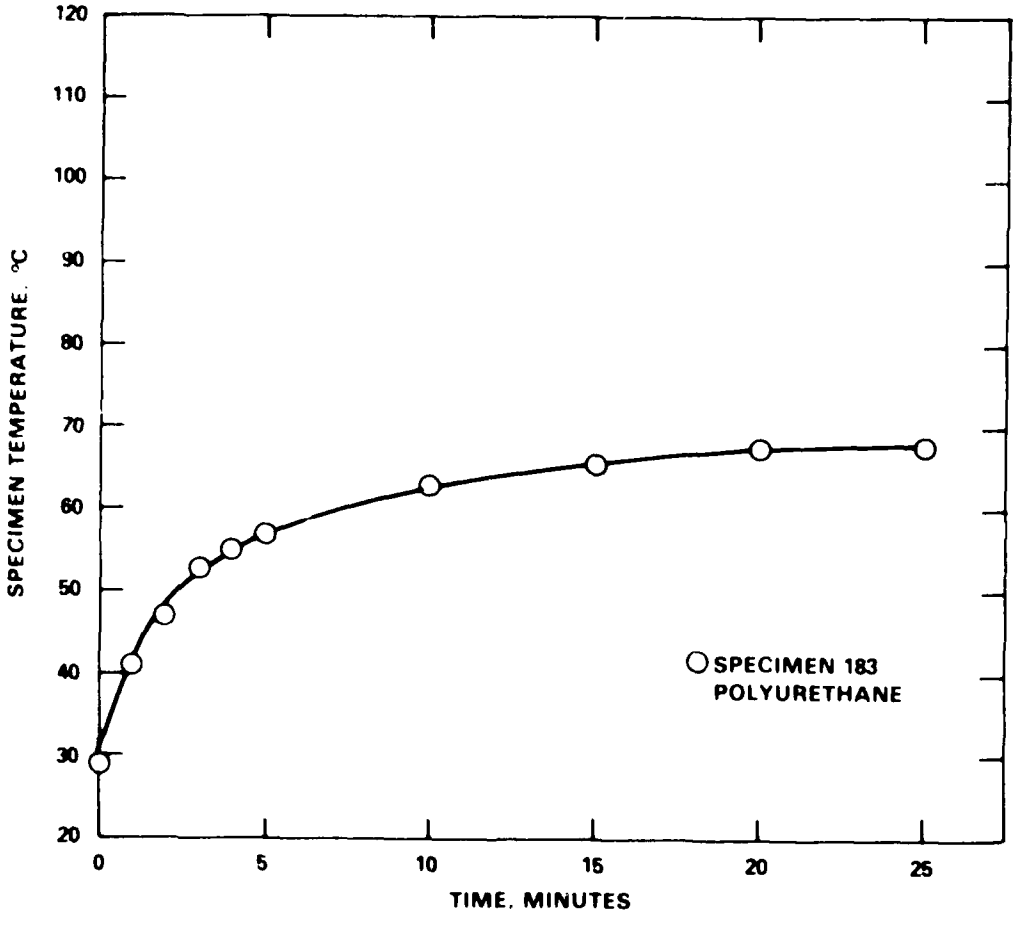

120
110
100
90
80
70
60
50
40
30
20
SPECIMEN TEMPERATURE °C
0
5
10
15
20
25
TIME, MINUTES
SPECIMEN ECB 502a
SPECIMEN ECB 502b

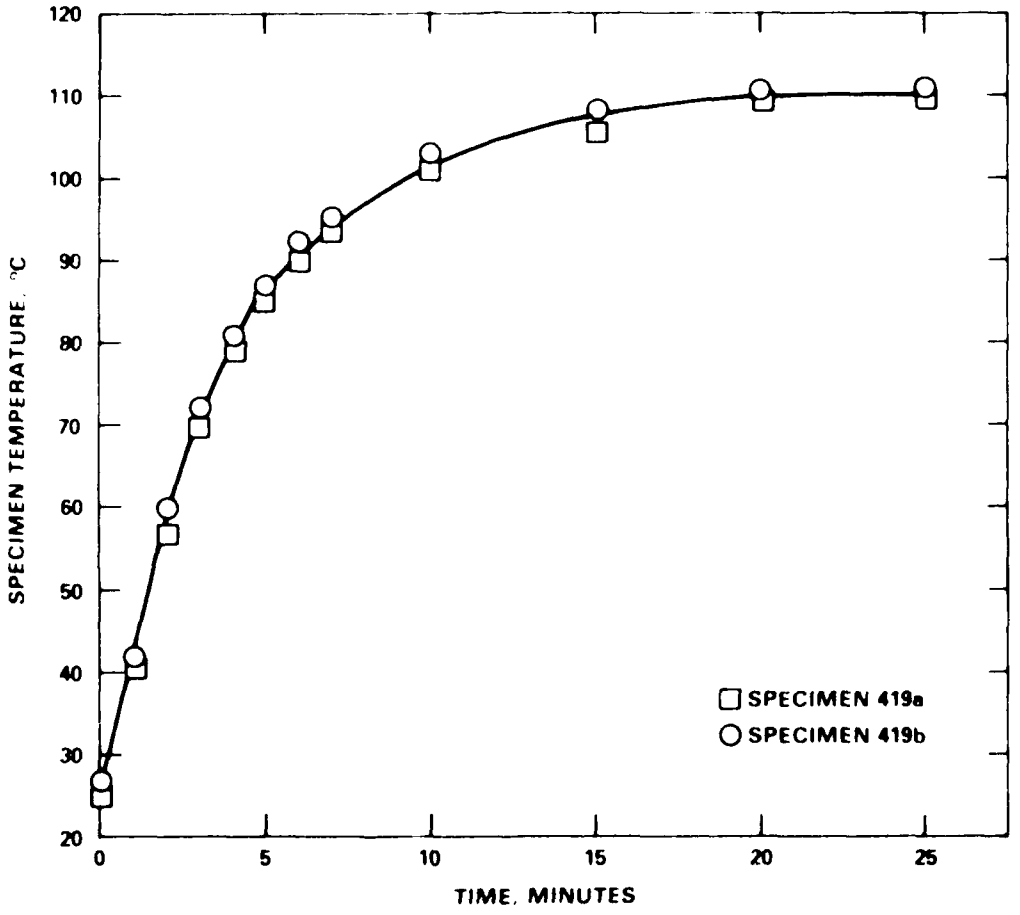

PU-205 NYLON FABRIC
PU-183 NYLON FABRIC
WEIGHT INCREASE (%)
SOAK TIME (DAYS)
0
1
2
3
4
5
6
0
2
4
6
8
10
12
16
18
20
22
24
26
28

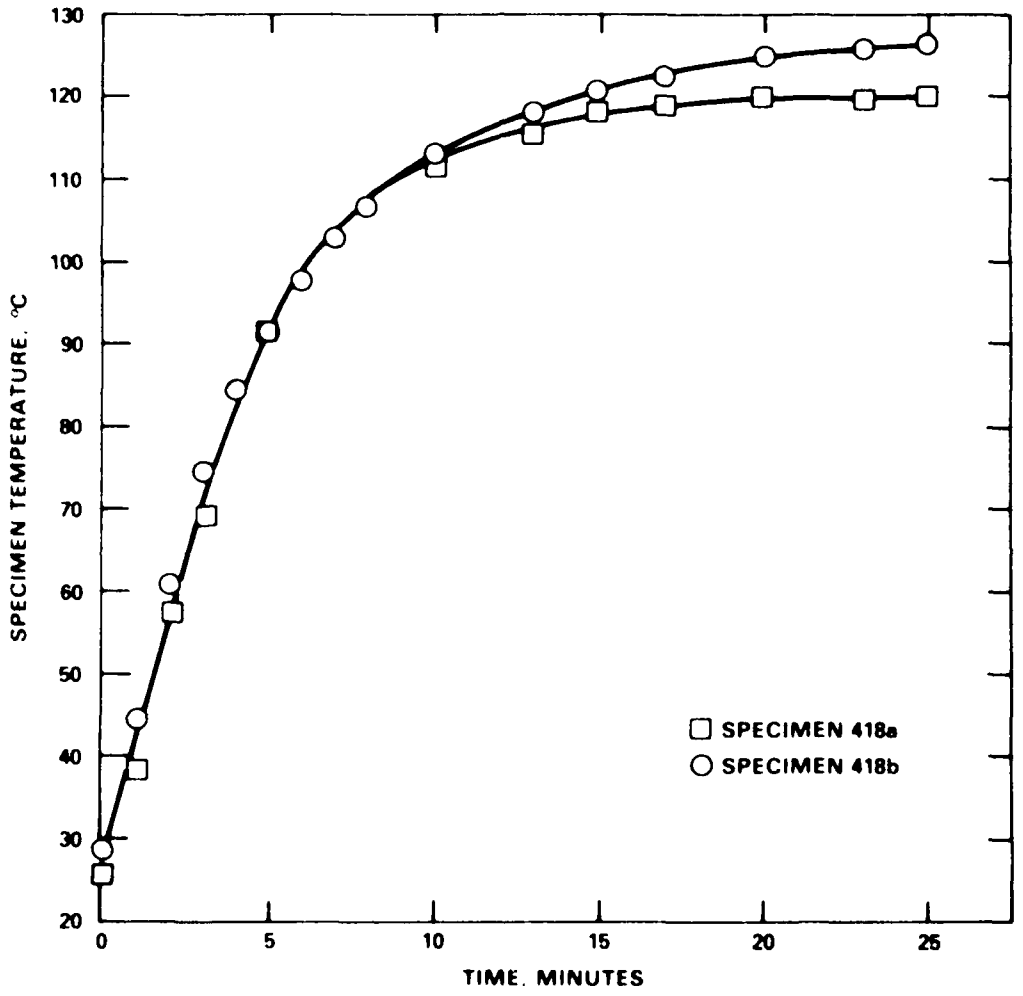

BAT 418 NYLON FABRIC
BAT 417 NYLON FABRIC
BAT 403 NYLON FABRIC
ECB-502 NYLON FABRIC
WEIGHT INCREASE (%)
0
2
4
6
8
10
12
14
16
18
0 10 20 30 40 50 60 70 80 90 100 110 120 130 140 150 160 170 180 190 200 210 220 230 240 250
DAYS SOAK IN SALT WATER

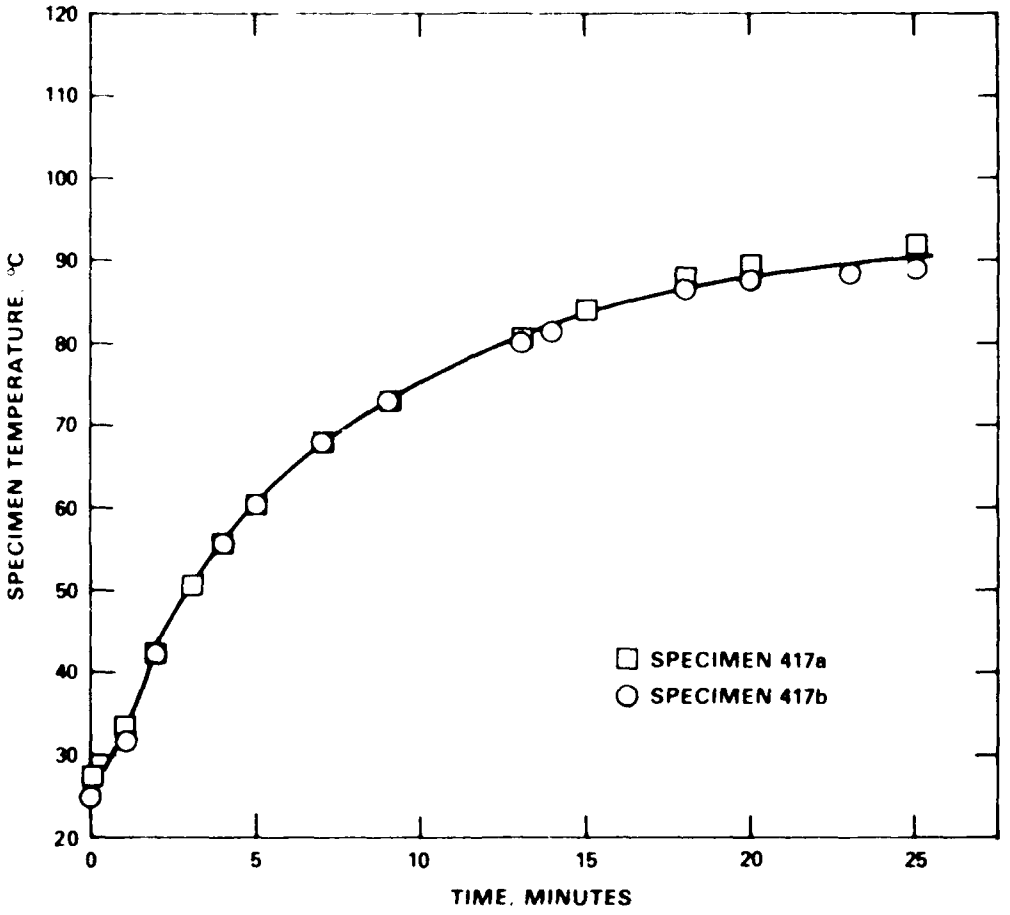

120
110
100
90
80
70
60
50
40
30
20
SPECIMEN TEMPERATURE, °C
0
5
10
15
20
25
TIME, MINUTES
SPECIMEN 206
POLYURETHANE


120
110
100
90
80
70
60
50
40
30
20
SPECIMEN TEMPERATURE, °C
0
5
10
15
20
25
TIME, MINUTES
SPECIMEN FAILED
(RUPTURED)
SPECIMEN 185
POLYURETHANE


120
110
100
90
80
70
60
50
40
30
20
SPECIMEN TEMPERATURE, °C
0
5
10
15
20
25
TIME, MINUTES
SPECIMEN 183
POLYURETHANE


120
110
100
90
80
70
60
50
40
30
20
0
5
10
15
20
25
SPECIMEN TEMPERATURE, °C
TIME, MINUTES
SPECIMEN 419a
SPECIMEN 419b


SPECIMEN TEMPERATURE, °C
TIME, MINUTES
SPECIMEN 418a
SPECIMEN 418b
130
120
110
100
90
80
70
60
50
40
30
20
0
5
10
15
20
25


120
110
100
90
80
70
60
50
40
30
20
SPECIMEN TEMPERATURE, °C
0
5
10
15
20
25
TIME, MINUTES
SPECIMEN 417a
SPECIMEN 417b

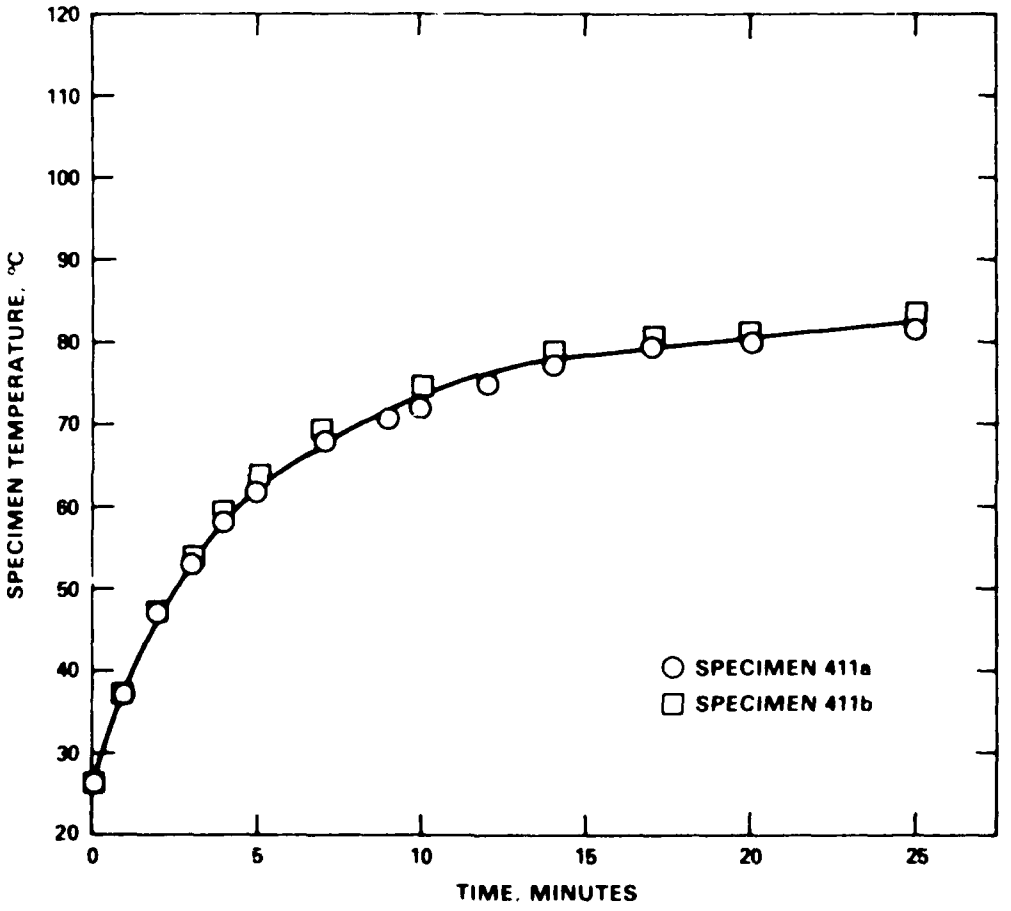
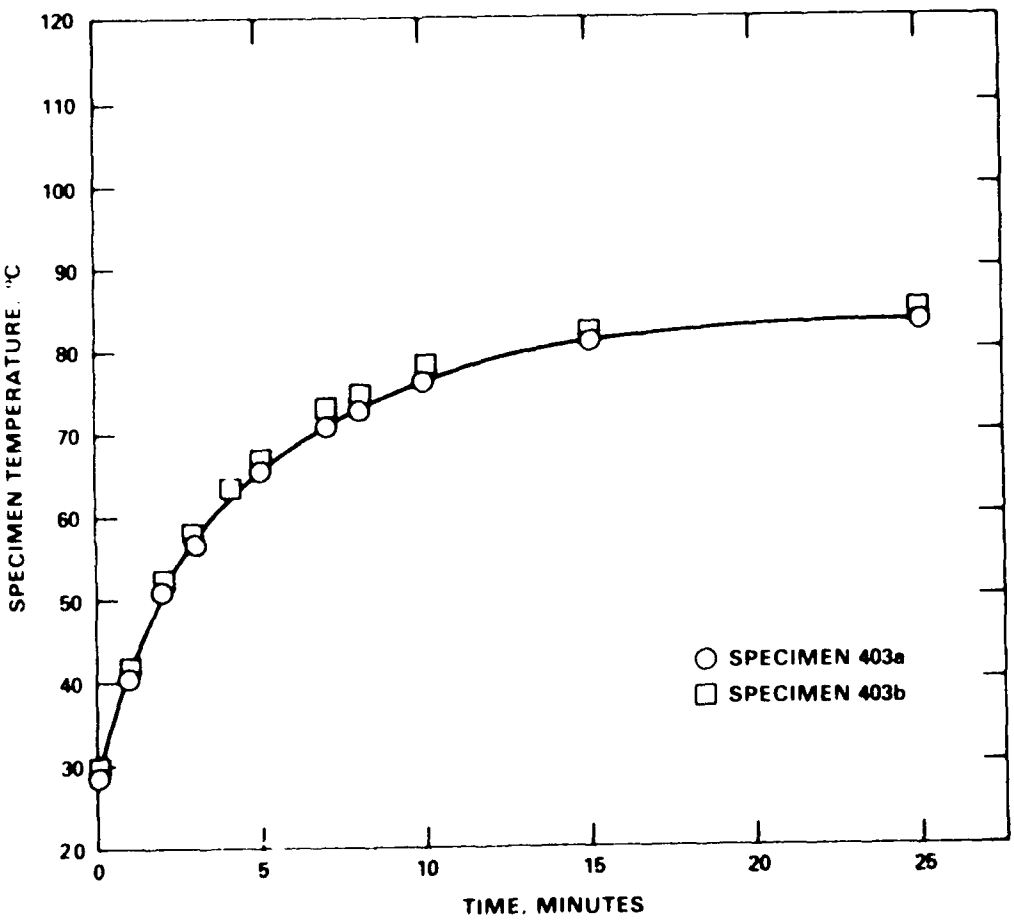
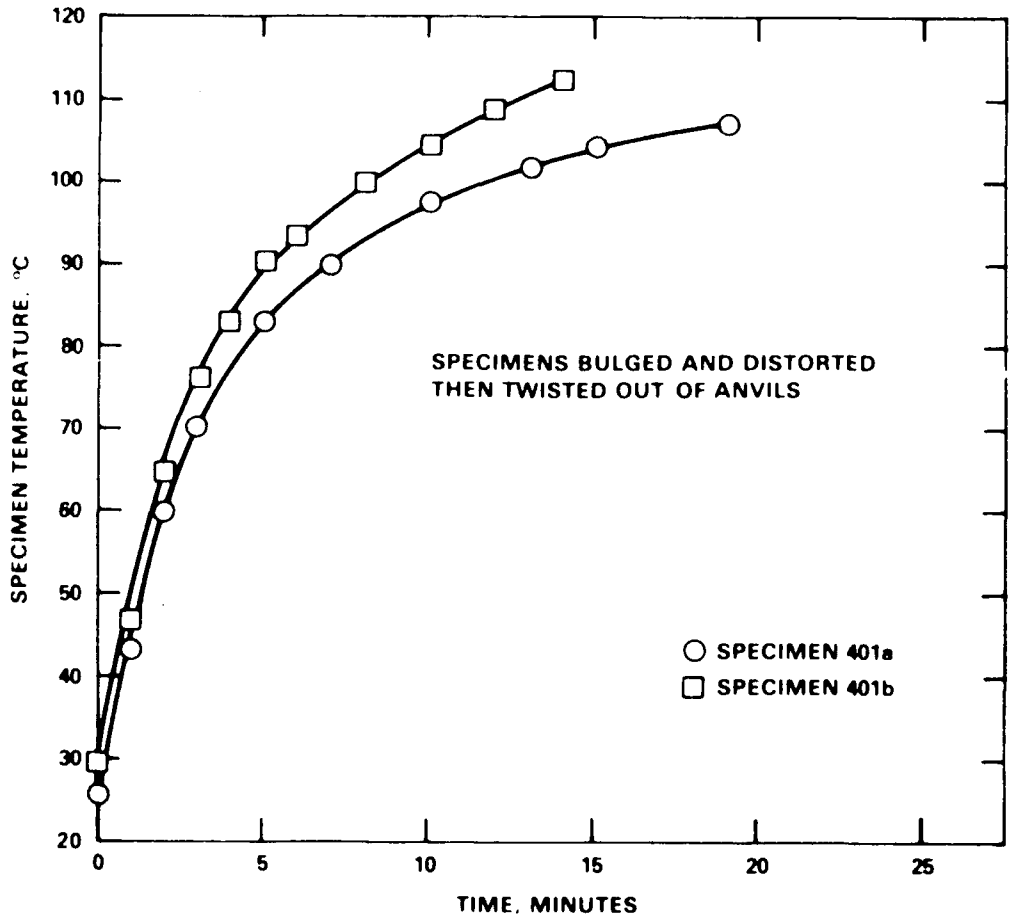


Figure 3-4. Heater-area temperature Rise During Cyclic Compressive Loading at 1400 Cycles per Minute Superimposed on a Compressive Load of 1.4 Pounds per Square Inch


120
110
100
90
80
70
60
50
40
30
20
SPECIMEN TEMPERATURE, °C
0
5
10
15
20
25
TIME, MINUTES
SPECIMEN 403a
SPECIMEN 403b

[illegible]


120
110
100
90
80
70
60
50
40
30
20
SPECIMEN TEMPERATURE, °C
0
5
10
15
20
25
TIME, MINUTES
SPECIMENS BULGED AND DISTORTED
THEN TWISTED OUT OF ANVILS
SPECIMEN 401a
SPECIMEN 401b

Hypalon Kevlar XA-22 Neoprene GNA Neoprene W Tetrone A Thionex	E.I. duPont de Nemours & Co., Inc. Wilmington, DE 19898
Antioxidant 702	Ethyl Corporation 451 Florida Street Baton Rouge, LA 70801
Wingstay 100	Goodyear Tire & Rubber Company Chemical Division 1485 E. Market Street Akron, OH 44316
STAN MAG Beads	Harwick Chemical Corporation 60 S. Seiberling Street P.O. Box 9360 Akron, OH 44305
Parel 58 Piccopale 100	Hercules Incorporated 910 Market Street Wilmington, DE 19899
T.12	M&T Chemical Incorporated Woodbridge Road P.O. Box 1104 Rahway, NJ 07065
Antioxidant MB Polycarbonate (Desmophen 2020)	Mobay Chemical Company Penn Lincoln Parkway West Pittsburgh, PA 15205
Santicizer 79TM Santicizer 711 Santocure Santocure NS Santoflex AW Santoguard PVI Sulfasan R	Monsanto Rubber Chemicals Division 260 Springside Drive Akron, OH 44313
Cumar MH	Neville Chemical Company Neville Island Pittsburgh, PA 15225
Kadox-15	New Jersey Zinc Company, Inc. Palmerton, PA 18071

[illegible]	Phillips Chemical Company A Division of Phillips Petroleum Company [illegible] Phillips Building Bartlesville, OK [illegible]
[illegible]	Polysar Limited [illegible] N. Front Street Sarnia, Ontario, CN N[illegible]M7
[illegible]	PPG Industries, Inc. Chemical Division One Gateway Center Pittsburgh, PA [illegible]
[illegible]	Rohm and Haas Company Independence Mall West Philadelphia, PA 19105
[illegible]	SEMCO, A Division of Products Research & Chemical Corporation [illegible]0 Jersey Avenue Gloucester City, NJ [illegible]
[illegible]	Shell Chemical Company One Shell Plaza Houston, TX [illegible]
[illegible] Rubber Brand Sulfur	Stauffer Chemical Company Nyala Farm Road Westport, CT 06880
[illegible] Light Oil [illegible] Oil [illegible] [illegible] II	Sun Refining and Marketing Company Lubes Division 1608 Walnut Street Philadelphia, PA 19103
[illegible]	Technical Processing, Inc. 136 Railroad Avenue Paterson, NJ 07509
Tillamox	Tulco, Inc. 9 Bishop Road Ayer, MA 01432

Carbowax Silane A-186 Silane A-1100 Silane A-1120	Union Carbide Corporation Silicones and Urethane Intermediates Division Old Ridgebury Road Danbury, CT 06817
E-S-E-N Flexzone 3C Rovalene 502 Sunproof Wax	Uniroyal Chemical Company Spencer Street Nagatuck, CT 06770
POP 88	UOP Incorporated Process Division 20 UOP Plaza Mt. Prospect & Algonquin Roads Des Plaines, IL 60016
Isonol 100	Upjohn Company Polymer Chemicals Division Battleground Road P.O. Box 685 LaPorte, TX 77571
AgeRite Hipar S AgeRite Resin D AgeRite Stalite AgeRite White Altax Cumate Ethyl Tuads Methyl Tuads	R.T. Vanderbilt Company, Inc. 30 Winfield Street Norwalk, CT 06855
Sunolite 666	Witco Chemical Corporation Organics Division 277 Park Avenue New York, NY 10017

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Mr. Meredith M. Schiegel

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1 Mr. William B. Cross
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1 Band, Lavis & Associates, Inc.
Mr. David R. Lavis

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1 Hover Systems Federal Corporation
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1 Hover Systems Incorporated
Mr. R. G. Longaker

1 Hercules Incorporated
Mr. A. L. Comen

1 Penn Engineering, Incorporated
Mr. Ray Lacey

1 Phillips Chemical Company
Dr. W. H. Cornell

1 The Firestone Tire and Rubber Co.
Mr. G. S. Kyker

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